: SEARCH REQUEST FORM

Scientific and Technical Information Center

	. 4	X	·		
Requester's Full Name: Camul	- Shumper	∿Examiner#:	79244 Da	ite: July 25, 2003	
Art Unit: 77 4 Phone N	lumber 30 5 44 8	Serial N	lumber: 10/0	77393	
Mail Box and Bldg/Room Location	. CP3 11 5 28 Resi	ults Format Pre	eferred (circle) PA	PER DISK E-MAIL	
	1 1			·	
If more than one search is subm			1 order of need. ******	*****	
Please provide a detailed statement of the		-	possible the subject	matter to be searched.	
Include the elected species or structures, k					
utility of the invention. Define any terms known. Please attach a copy of the cover s			mples or relevant cita	itions/authors, etc, if	
- V		1.	millichi.		
Title of Invention: Multilayer u Spaurs, touch Street					
Inventors (please provide full names):	Charles C: C	Indusm	', Ronald	5. Coke;	
~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	edman				
. <u>J 2 </u>	Misham		Najar."		
Earliest Priority Filing Date:	2117 (2007				
For Sequence Searches Only Please include appropriate serial number.					
appropriate serial number. Please do (Jumellas		,		- 11 , H	
	Canada I	~ 0	aims 1-	- 41 WUR	
Pound do	Jeanen "	n Cr.			
para or		•	V.		
	· 1. /.	./			
/ smiller	MALLIACE	1.	· / .		
portación	Months of the second				
	-	~ 1	1/		
		1 Ma	Mo	*	
		Ü			
	Parameter (1)		-		
% .		, .		y	
The second of th	ŕ				
		•	Same of the same o		
	•			<i>:</i>	
		•	•	× x	
				•	
	_. به در _{در}	•			
				•	
The state of the s	•	-			
	******		. 1		
STAFF USE ONLY	Type of Search	Vend	ors and cost where	applicable	
Searcher:	NA Sequence (#)	STN	274.38		
Searcher Phone #:	AA Sequence (#)	Dialog	7		
	1 (1	\			
Searcher Location:	Structure (#)	- Questel/Orbit			
Date Searcher Picked Up:	Bibliographic /	Dr.Link			
Date Completed: (-30-03	Litigation	Lexis/Nexis			
Searcher Prep & Review Time:	Fulltext	Sequence Systems	s	<u> </u>	
Glerical Prep Times	Patent Family	WWW/Internet _	\(\frac{1}{1}\)		
Online Time: 130	Other	Other (specify)	4 /		
PTO 1500 (8 01)		3	W 1 -		
PTO-1590 (8-01)		7. 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		

=> file reg FILE 'REGISTRY' ENTERED AT 12:36:51 ON 30 JUL 2003 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2003 American Chemical Society (ACS)

=> display history full l1-

L1	FILE 'LREG	ISTRY' ENTERED AT 10:48:28 ON 30 JUL 2003 STR
L2 L3 L4	18	STRY' ENTERED AT 10:51:22 ON 30 JUL 2003 SCR 2043 SEA SSS SAM L1 AND L2 SEA SSS FUL L1 AND L2 SAV L4 THO393/A
L5		ENTERED AT 11:05:01 ON 30 JUL 2003 SEA (FILM? OR THINFILM? OR LAYER? OR OVERLAY? OR OVERLAID? OR LAMIN? OR LAMEL? OR SHEET? OR LEAF? OR FOIL? OR COAT? OR TOPCOAT? OR OVERCOAT? OR VENEER? OR SHEATH? OR COVER? OR ENVELOP? OR ENCAS? OR ENWRAP? OR OVERSPREAD?)/BI,AB
L6		ENTERED AT 11:14:16 ON 30 JUL 2003 SEA LAMIN? OR LAMEL? OR MULTILAYER? OR MULTICOAT? OR MULTIFILM? OR (MULTI OR MULTIPL? OR PLURAL? OR NUMEROUS? OR MANY OR MULTITUD? OR MANIFOLD? OR MULTIFOLD? OR SEVERAL OR FEW) (2A) (L5 OR CLAD?)
L7 L8	77835	SEA L4 SEA (ELECTROLUM!N? OR ORGANOLUM!N? OR (ELECTRO OR ORGANO OR ORG#)(2A)LUM!N? OR LIGHT?(2A)(EMIT? OR EMISSION?) OR EL OR E(W)L OR L(W)E(W)D OR OLED)/BI,AB OR LED/IT
L9	40019	SEA OLED OR O(W)L(W)E(W)D OR TOUCHSCREEN? OR TOUCHPANEL? OR TOUCHDISPLAY? OR TOUCH?(2A) (SCREEN? OR PANEL? OR DISPLAY?) OR DISPLAY?(2A) (DEVICE? OR EQUIP? OR APPARAT? OR APP## OR ASSEMBLY OR ASSEMBLIES OR SCREEN? OR PANEL? OR MONITOR?) OR CRT OR C(W)R(W)T OR CATHOD##(2A)RAY#(2A)TUBE#
L10	169	SEA L7 AND L6
L11		SEA L10 AND L8
L12		SEA L10 AND L9
L13		SEA TOUCHSCREEN? OR TOUCHPANEL? OR TOUCH? (2A) (SCREEN? OR PANEL? OR DISPLAY?)
L14		SEA L7 AND L13
L15	1	SEA L10 AND SPACER#
L16	FILE 'LCA'	ENTERED AT 11:30:05 ON 30 JUL 2003 SEA SPINCOAT? OR WEBCOAT? OR SPRAYCOAT? OR ELECTROCOAT?

OR (SPIN? OR SPUN? OR WEB OR WEBS OR WEBBED OR WEBBING#

OR SPRAY? OR ELECTRO#)(2A)COAT?

```
L17
             53 SEA MICROSPHER? OR MICROBALL# OR MICROGLOB? OR NANOSPHER?
                 OR NANOBALL# OR MICROBALL? OR NANOSPHER? OR NANOBALL#
                OR NANOGLOB? OR (MICRO OR NANO) (2A) (SPHER? OR BALL# OR
                GLOB?)
            120 SEA ENCAPSUL? OR MICROENCAPSUL? OR MICROCAPSUL? OR
L18
                CAPSULAT? OR NANOENCAPSULAT? OR NANOCAPSUL?
     FILE 'HCA' ENTERED AT 12:16:39 ON 30 JUL 2003
             11 SEA L10 AND L16
L19
              0 SEA L10 AND L17
L20
            1 SEA L10 AND L18
L21
             7 SEA L7 AND SPACER#
L22
             16 SEA L11 AND L12
L23
             10 SEA L14 OR L15 OR L21 OR L22
L24
             24 SEA (L19 OR L23) NOT L24
L25
L26
             4 SEA L12 NOT (L24 OR L25)
             28 SEA (L19 OR L23 OR L12) NOT L24
L27
            49 SEA L11 NOT (L24 OR L27)
L28
                E COATING PROCESS/CV
         105160 SEA "COATING PROCESS"/CV
                E COATING MATERIALS/CV
         235874 SEA "COATING MATERIALS"/CV
L30
             1 SEA L28 AND (L29 OR L30)
L31
             29 SEA L27 OR L31
L32
             37 SEA L28 AND (1907-2001/PY OR 1907-2001/PRY)
L33
             10 SEA L24 AND (1907-2002/PY OR 1907-2002/PRY)
L34
L35
             27 SEA L32 AND (1907-2002/PY OR 1907-2002/PRY)
```

FILE 'REGISTRY' ENTERED AT 12:36:51 ON 30 JUL 2003

```
=> d 14 que stat
L1 STR

7 O 0 8
C C C 4
5
```

NODE ATTRIBUTES:
CONNECT IS E2 RC AT 5
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 7

STEREO ATTRIBUTES: NONE L2 SCR 2043

L4 352 SEA FILE=REGISTRY SSS FUL L1 AND L2

100.0% PROCESSED 4090 ITERATIONS SEARCH TIME: 00.00.01

352 ANSWERS

=> file hca FILE 'HCA' ENTERED AT 12:38:44 ON 30 JUL 2003 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2003 AMERICAN CHEMICAL SOCIETY (ACS)

=> d 124 1-10 cbib abs hitstr hitind

ANSWER 1 OF 10 HCA COPYRIGHT 2003 ACS on STN L24139:45127 Material for making a conductive pattern. Lamotte, Johan; Louwet, Frank; Cloots, Tom; Van Aert, Huub (Agfa-Gevaert, Belg.). PCT Int. Appl. WO 2003050824 A1 20030619, 47 pp. DESIGNATED STATES: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, W: CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM; RW: AT, BE, BF, BJ, CF, CG, CH, CI, CM, CY, DE, DK, ES, FI, FR, GA, GB, GR, IE, IT, LU, MC, ML, MR, NE, NL, PT, SE, SN, TD, TG, TR. (English). CODEN: PIXXD2. APPLICATION: WO 2002-EP13159 20021122. PRIORITY: EP 2001-739 20011211.

AΒ A material contg. an intrinsic conducting polymer for making a conductor pattern without the use of hazardous materials. A material for making a conductive pattern, the material comprising a support and a heat-differentiable element, the heat-differentiable element comprising an outermost layer contg. a polyanion and an intrinsically conductive polymer and optionally a 2nd layer contiguous with the outermost layer, characterized in that the outermost layer and/or the optional 2nd layer contains hydrophobic thermocoagulable latex particles in a wt. ratio range with respect to the intrinsically conductive polymer in the range of 20:1 to 1:5. The hydrophobic thermocoagulable latex particles are capable upon heating of increasing the cond. of the heated parts of the outermost layer relative to the unheated parts of the outermost layer and/or changing the removability of the heated parts of the outermost layer relative to the unheated parts of the outermost layer and the heat-differentiable element does not comprise a di- or polyhydroxy org. compd. or an aprotic compd. with a dielec. const. .gtoreq. 15; a method of making a conductive pattern on a support therewith; and a use of the material for making a conductive pattern in making an

electronic circuit in the prodn. of an elec. or semiconductor device such as a printed circuit board, an integrated circuit, a display or touch screen, an

electroluminescent device or a photovoltaic cell.

IT 126213-51-2P, PEDOT

(material contg. intrinsic conducting polymer for making conductor pattern without use of hazardous materials)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

IC ICM H01B001-12

CC 76-2 (Electric Phenomena) Section cross-reference(s): 38, 74

IT 126213-51-2P, PEDOT

(material contg. intrinsic conducting polymer for making conductor pattern without use of hazardous materials)

L24 ANSWER 2 OF 10 HCA COPYRIGHT 2003 ACS on STN

138:212615 Method for manufacturing low cost electroluminescent (EL) illuminated membrane switches. Stevenson, William C.; Lau, James L. (Novatech Electroluminescent, Inc., USA). U.S. Pat. Appl. Publ. US 2003041443 A1 20030306, 19 pp. (English). CODEN: USXXCO. APPLICATION: US 2001-942339 20010830.

A method of fabricating low cost electroluminescent (EL) illuminated ABmembrane switches is described that entails die cutting, embossing or chem. etching the metal foil surface of a metal foil bonded, light transmitting flexible elec. insulation to simultaneously form .qtoreq.1 front capacitive electrodes, membrane switch contacts and elec. shunt, elec. distribution means and elec. terminations that together comprise a flexible printed circuit panel, coupling the circuit substrate to a precisely positioned indexing system, coating the front metal foil capacitive electrodes with a light transmissive elec. conductive layer, applying a layer of electroluminescent phosphor to the elec. conductive layer, applying a layer of capacitive dielec. layer to the phosphor layer, applying a rear capacitive electrode over the capacitive dielec. layer, thus forming an electroluminescent lamp portion, applying a transparent dielec. coating to the entire surface of the lamp and substrate with open portions exposing elec. terminations, switch contacts and shunt, applying a spacer to surround the switch shunt, providing

an isolation barrier, and applying an intermediary material to the surface of the isolated rear EL electrode thus forming a switch actuator. Finally, the illuminated switch pattern is die-cut from the substrate material, and is then folded into 3 layers forming the final illuminated membrane switch.

IT 126213-51-2, PEDOT

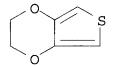
(light transmissive electrode; method of fabricating electroluminescent membrane switches)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



IC ICM H05B033-10

ICS H05K003-10; H01B013-00

NCL 029622000; 216005000; 216013000; 216024000; 438099000; 029847000; 029846000

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT 126213-51-2, PEDOT

(light transmissive electrode; method of fabricating electroluminescent membrane switches)

L24 ANSWER 3 OF 10 HCA COPYRIGHT 2003 ACS on STN

138:97572 Electroluminescence properties of an alternating blue-green light-emitting copolymer consisting of soft and rigid segments. Wang, Hai-qiao; Li, Xiao-yu (The Key Lab. Sci. Technology Controllable Chem. Reactions, Ministry Education, Sch. Materials Sci. Eng., Beijing Univ. Chemical Technology, Beijing, 100029, Peop. Rep. China). Gongneng Gaofenzi Xuebao, 15(3), 276-280 (Chinese) 2002. CODEN: GGXUEH. ISSN: 1004-9843. Publisher: Gongneng Gaofenzi Xuebao Bianjibu.

AB A blue-green light-emitting copolymer (TEO-NV) contg. alternating 1,5-(3,5-dimethyloxy styrene) naphthalene as chromophore and tri(ethylene oxide) as functional **spacer**, was synthesized. Its chem. structure was characterized and luminescent properties was investigated. Thermal properties were measured with DSC and TGA under nitrogen atm. TEO-NV has excellent thermal stability and the decompn. temp. is high up to 409 .degree.C with Tg = 42 .degree.C. TEO-NV can be sol. in many org. solvents, such as chloroform, methylene dichloride and toluene, and the polymer soln. can be

spin-coated onto various substrates giving highly transparent and homogeneous thin film. TEO-NV is a typical blue-green light-emitting copolymer with a max. EL emitting peaks at 499 nm. A light-emitting diode (LED) based on TEO-NV was successfully fabricated. Its threshold voltage was ca. 5 V for light emission, and the max. brightness was 295 cd/m2 at forward bias 20 V. 126213-51-2

IT 126213-51-2

(for electroluminescent devices made of alternating blue-green light-emitting copolymer consisting of soft and rigid segments)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 36

TT 7429-90-5, Aluminum, uses 7440-70-2, Calcium, uses 50851-57-5
50926-11-9, Indium tin oxide 126213-51-2

(for electroluminescent devices made of alternating blue-green light-emitting copolymer consisting of soft and rigid segments)

L24 ANSWER 4 OF 10 HCA COPYRIGHT 2003 ACS on STN

- 137:279555 A luminescent copolymer containing PPV-based chromophores and flexible tri(ethylene oxide) spacers. Wang, Haiqiao; Sun, Qingjiang; Li, Yongfang; Liu, Deshan; Wang, Xiaogong; Li, Xiaoyu (The Key Laboratory of Science and Technology of Controllable Chemical Reactions, School of Materials Science and Engineering, Ministry of Education, Beijing University of Chemical Technology, Beijing, 100029, Peop. Rep. China). Reactive & Functional Polymers, 52(2), 61-69 (English) 2002. CODEN: RFPOF6. ISSN: 1381-5148. Publisher: Elsevier Science B.V..
- AB A luminescent triethylene oxide-phenylene vinylene block copolymer (TEO-MPV) was synthesized through Wittig polycondensation reaction. The structure of the copolymer was verified using FTIR, 1H NMR, and elemental anal. The electrochem. properties of the copolymer were evaluated and the HOMO and LUMO energy levels of the copolymer were estd. by cyclic voltammetry. Thermal anal. showed that the glass transition temp. (Tg) of the copolymer is about 85.6.degree. and the decompn. temp. is over 384.degree.. The fluorescence quantum yield of TEO-MPV chloroform soln. reaches 99.05%, much higher than that of analogous polymers and has greenish-blue emission. An

ITO/TEO-MPV/Al single layer LED assembly , ITO/PEDOT-PSS/TEO-MPV/Ca (Al) bilayer LED, and a light-emitting electrochem. cell (LEC) were fabricated. The LEC devices have lower turn-on and operating voltage than corresponding LED devices.

IT **126213-51-2**, PEDOT

(carrier layer; Wittig polycondensation in prepn. of luminescent poly(phenylene vinylene-ethylene oxide) and electrochem. and luminescence and performance as emitter layer in devices)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CC 35-5 (Chemistry of Synthetic High Polymers) Section cross-reference(s): 36, 73

9003-53-6D, Polystyrene, sulfonated 126213-51-2, PEDOT (carrier layer; Wittig polycondensation in prepn. of luminescent poly(phenylene vinylene-ethylene oxide) and electrochem. and luminescence and performance as emitter layer in devices)

L24 ANSWER 5 OF 10 HCA COPYRIGHT 2003 ACS on STN

137:239513 Polymer organic light emitting device with improved color control. Roitman, Daniel B.; Antoniadis, Homer (Agilent Technologies, Inc., USA). Eur. Pat. Appl. EP 1244153 A2 20020925, 7 pp. DESIGNATED STATES: R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR. (English). CODEN: EPXXDW. APPLICATION: EP 2001-130337 20011219. PRIORITY: US 2001-814381 20010321.

An org. light emitting device (OLED) for emitting light at a predetd. wavelength is described comprising an anode; a cathode layer; an electroluminescent layer comprising an org. light emitting compd. for generating light including light of the predetd. wavelength by the recombination of holes and electrons, the electroluminescent layer being located between the anode and cathode and being connected to them; and a first and a second reflector, the second reflector being partially reflecting and being displaced from the first reflector, such that the optical path length between the first and second reflectors is D; a spacer layer comprising a material that is transparent at .lambda., wherein D = N.lambda./2 (N = pos. integer). The spacer may include a hole transport material located between the electroluminescent layer and the anode layer.

IT 126213-51-2, PEDOT

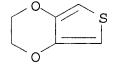
(light emitting layer; polymer org. light emitting device with improved color control)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



IC ICM H01L051-20

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
Section cross-reference(s): 76

ST org light emitting device color control spacer layer

IT 126213-51-2, PEDOT

(light emitting layer; polymer org. light emitting device with improved color control)

L24 ANSWER 6 OF 10 HCA COPYRIGHT 2003 ACS on STN

137:171129 Antisoiling hardcoats containing inorganic oxides and perfluoropolyethers. Liu, Junkang J.; Fong, Bettie C.; Kluge, Bruce D. (USA). U.S. Pat. Appl. Publ. US 2002114934 A1 20020822, 20 pp., Cont.-in-part of U.S. Ser. No. 633,835. (English). CODEN: USXXCO. APPLICATION: US 2001-923749 20010807. PRIORITY: US 2000-633835 20000807.

AB An antisoiling hardcoated film comprises a substantially transparent substrate, a hardcoat layer comprising inorg. oxide particles dispersed in a binder matrix, and an antisoiling layer comprising a perfluoropolyether. The antisoiling layer can be very thin, thus reducing the cost of the perfluoropolyether. The film has very good scratch, smudge and glare resistance and very good interlayer adhesion and durability. The film can be in the form of a single flexible substrate or a stack of such substrates. The film or stack can be sized to fit the display screen of an electronic display device such as a personal digital assistant or cell phone.

IT 155090-83-8, BAYTRON P

(antisoiling hardcoats contg. inorg. oxides and perfluoropolyethers)

RN 155090-83-8 HCA

CN Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

```
CM 1

CRN 126213-51-2

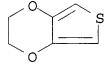
CMF (C6 H6 O2 S) x

CCI PMS

CM 2

CRN 126213-50-1

CMF C6 H6 O2 S
```



CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS

D1-CH-CH2

D1-SO3H

IT 155090-83-8, BAYTRON P
(antisoiling hardcoats contg. inorg. oxides and perfluoropolyethers)

ANSWER 7 OF 10 HCA COPYRIGHT 2003 ACS on STN L24 136:94638 Making encapsulated organic electronic devices. McCormick, Fred B.; Baude, Paul F.; Vernstrom, George D. (3M Innovative Properties Company, USA). PCT Int. Appl. WO 2002005361 A1 20020117, 33 pp. DESIGNATED STATES: W: AE, AG, AL, AM, AT, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, CZ, DE, DE, DK, DK, DM, DZ, EE, EE, ES, FI, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM; RW: AT, BE, BF, BJ, CF, CG, CH, CI, CM, CY, DE, DK, ES, FI, FR, GA, GB, GR, IE, IT, LU, MC, ML, MR, NE, NL, PT, SE, SN, TD, TG, TR. (English). CODEN: PIXXD2. APPLICATION: WO 2000-US31393 20001115. PRIORITY: US 2000-614993 20000712.

The invention provides methods for making encapsulated AΒ org. electronic devices (OED) including org. LEDs (OLED). The present invention can provide a robust OED device by means of in situ edge sealing enhancing structural integrity and device lifetime. The edge sealing is provided by using an adhesive component applied to a substrate prior to OED element deposition. thin layer of an adhesive (pressure sensitive adhesive, hot melt, or curable) is applied to release liner, openings are cut in the adhesive/liner composite, then the composite is adhered to an electrode-coated substrate. Alternatively, an adhesive may be applied directly onto the electrode-coated substate, e.g., by printing in a desired pattern, optionally partially cured or dried, then covered with .gtoreq.1 liners that act as a mask during deposition of the OLED elements. Another method would be to prep. a blank liner with a patterned adhesive, then die cut openings complementary to the adhesive pattern in the liner to allow deposition of OLED elements once the adhesive/liner is placed on the substrate.

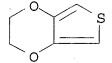
IT 126213-51-2, Poly(ethylenedioxythiophene)
(conductive polymer in light emitting construction; making encapsulated org. electronic devices)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



```
ICM H01L051-20
IC
     ICS H01L051-40; H05B033-04
     76-3 (Electric Phenomena)
CC
     Section cross-reference(s): 38
     encapsulated org electronic device prepn
ST
     Electron beam evaporation
IT
        (alumina layer; making encapsulated org. electronic
        devices)
IT
     Siloxanes (nonpolymeric)
        (coated on metal foil as adhesive-coated liner; making
        encapsulated org. electronic devices)
ΙT
     Polyesters, processes
        (coated with siloxanes or fluorocarbons as adhesive-coated liner;
        making encapsulated org. electronic devices)
IT
     Adhesives
        (conductive, thermal, elec.; making encapsulated orq.
        electronic devices)
IT
     Adhesives
        (curable; making encapsulated org. electronic devices)
IT
     Fluoropolymers, processes
        (film as adhesive-coated liner; making encapsulated
        org. electronic devices)
IT
    Hydrocarbons, processes
        (fluoro, coated on metal foil as adhesive-coated liner; making
        encapsulated org. electronic devices)
IT
     Electric contacts
     Electroluminescent devices
       Encapsulation
     Lamps (nonelectric)
    Microelectronic devices
     Release coatings
     Sealing
     Shadow masks
        (making encapsulated org. electronic devices)
IT
     Foils
        (metal as protective layer; making encapsulated org.
        electronic devices)
IT
    Multilayers
        (polymer as protective layer; making encapsulated org.
        electronic devices)
IT
        (polymeric; making encapsulated org. electronic
        devices)
IT
     Adhesives
        (pressure-sensitive; making encapsulated org.
```

electronic devices)

- IT Glass, uses
 - (thin flexible as protective layer; making encapsulated org. electronic devices)

- IT 1314-13-2, Zinc oxide, uses 7429-90-5, Aluminum, uses 7439-93-2, 7439-95-4, Magnesium, uses 7440-19-9, Samarium, Lithium, uses 7440-22-4, Silver, uses 7440-39-3, Barium, uses 7440-65-5, Yttrium, uses 7440-70-2, 7440-57-5, Gold, uses Calcium, uses 50926-11-9, ITO 53740-87-7 (counter electrode; making encapsulated org. electronic devices)
- IT 1344-28-1, Alumina, processes (electron beam evapn. on ITO; making encapsulated org. electronic devices)
- IT 1332-29-2, Tin oxide (fluorine, counter electrode; making encapsulated org. electronic devices)
- IT 147-14-8, Copper phthalocyanine 155306-71-1, C545T (in light emitting construction; making encapsulated org. electronic devices)
- IT 2085-33-8, Tris(8-hydroxyquinolinato)aluminum (light emitting layer; making encapsulated org. electronic devices)
- L24 ANSWER 8 OF 10 HCA COPYRIGHT 2003 ACS on STN
- 132:140918 Flexible glass substrates and laminated light-emitting devices based on this substrate. Krijn, Marcellinus P. C. M.; Dona, Marinus J. J.; Swinkels, Johannes M. M.; Vleggaar, Jeroen J. M. (Koninklijke Philips Electronics N.V., Neth.). PCT Int. Appl. WO 2000005180 A1 20000203, 12 pp. DESIGNATED STATES: W: JP, KR; RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE. (English). CODEN: PIXXD2. APPLICATION: WO 1999-EP4759 19990707. PRIORITY: EP 1998-202421 19980720.
- The substrate comprises a glass sheet with a thickness .ltoreq.0.1 mm (e.g., 60-80 .mu.m) being provided with a layer of a synthetic resin having a thickness less or equal of the glass sheet thickness. This substrate is flexible and can be bent to a 1 cm curvature radius without fracturing. The light-emitting device comprises a stack of two substrates and spacers between them, where the space is filled with a mixt. of inert gas with mercury vapor.

The substrate may be used as flexible light source in (1) light-emitting devices for liq.-crystal displays, such as poly-LED or Plasma Addressed Liq. Crystal displays, and (2) electrochromic devices.

IT 126213-51-2

(flexible glass substrates and laminated light-emitting devices based on this substrate)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

IC ICM C03C017-32

ICS G02F001-133

CC 57-1 (Ceramics)

Section cross-reference(s): 74

ST flexible laminated glass substrate liq crystal display

IT Electrochromic devices Liquid crystal displays

(flexible glass substrate and laminated light-emitting device based on this substrate)

IT Sheet glass

(flexible glass substrate and laminated light-emitting device based on this substrate)

IT Noble gases, uses

(mixt. with mercury vapor; flexible glass substrate and laminated light-emitting device based on this substrate)

IT Plastics, uses

(polyurethane; flexible glass substrate and laminated light-emitting device based on this substrate)

IT 1067-53-4, Z-6082

(adhesive; flexible glass substrates and laminated light-emitting devices based on this substrate)

TT 7440-22-4, Silver, uses 7440-50-8, Copper, uses (component of bonding layer; flexible glass substrates and laminated light-emitting devices based on this substrate)

IT 7440-64-4, Ytterbium, uses

(component of cathode layer; flexible glass substrates and laminated light-emitting devices based on this substrate)

IT 96352-95-3P, NeoRez R-970 154214-68-3P, NeoRez R-972 256949-56-1P, NeoRez R 980

(flexible glass substrates and laminated light-emitting

devices based on this substrate)

IT 126213-51-2

(flexible glass substrates and laminated light-emitting devices based on this substrate)

IT 12788-05-5, Indium-tin eutectic

(moisture-proofing layer on the glass; flexible glass substrates and laminated light-emitting devices based on this substrate)

IT 138184-36-8, MEH-PPV

(spun layer; flexible glass substrates and laminated light-emitting devices based on this substrate)

IT 7439-97-6, Mercury, uses

(vapor, mixt. with inert gas; flexible glass substrates and laminated light-emitting devices based on this substrate)

L24 ANSWER 9 OF 10 HCA COPYRIGHT 2003 ACS on STN

- 131:323230 Synthesis and characterization of oligo- and crown ether-substituted polythiophenes a comparative study. Scheib, Stefan; Bauerle, Peter (Abteilung Organische Chemie II, Universitat Ulm, Ulm, D-89081, Germany). Journal of Materials Chemistry, 9(9), 2139-2150 (English) 1999. CODEN: JMACEP. ISSN: 0959-9428. Publisher: Royal Society of Chemistry.
- The synthesis of two series of thiophenes substituted with crown and AB oligoether groups either via isolating oxaalkyl chains or in direct .pi.-conjugation is described. Electrooxidative polymn. leads to the corresponding crown and oligoether-functionalized polythiophenes. Their electrochem. and spectroscopic properties depend on the length of the spacer and the type of the ether unit. The polymers reveal a high mean conjugation. specific and strong influence of alkali ions on the electrochem. behavior is found for several polymers. The selectivities correspond to the match of the cation size without solvent shell and the inner diam. of the crown ether units. Spectroelectrochem. expts. corroborate that the changes in redox properties are due to a hindered diffusion of the counter anions into the film when the polymer is oxidized. Due to the structural variation novel materials sensitive to different cations are obtained. Importantly, in these conjugated polymers chem. information which corresponds to a selective host-guest interaction of the alkali metal cations and the ether units is transduced into the change of an elec. signal.

IT 163657-78-1P 249513-23-3P

(prepn. and characterization of)

RN 163657-78-1 HCA

CN Thieno[3,4-b]-1,4,7,10,13-pentaoxacyclopentadecin, 2,3,5,6,8,9,11,12-octahydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 122372-74-1 CMF C12 H18 O5 S

RN 249513-23-3 HCA

CN Thieno[3,4-b]-1,4,7,10-tetraoxacyclododecin, 2,3,5,6,8,9-hexahydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 122372-73-0 CMF C10 H14 O4 S

CC 37-3 (Plastics Manufacture and Processing)

IT 119889-86-0P 157483-26-6P **163657-78-1P** 249513-18-6P

249513-19-7P 249513-20-0P 249513-21-1P 249513-22-2P

249513-23-3P

(prepn. and characterization of)

L24 ANSWER 10 OF 10 HCA COPYRIGHT 2003 ACS on STN

129:261150 Odd-even effect in the association of chiral

poly(3,4-dialkoxythiophenes). Ramos Lermo, M. E.; Langeveld-Voss, B. M. W.; Meijer, E. W. (Lab. Macromol. and Organic Chem., Eindhoven Univ. Technol., Eindhoven, 5600 MB, Neth.). Polymer Preprints (American Chemical Society, Division of Polymer Chemistry), 39(2), 1087-1088 (English) 1998. CODEN: ACPPAY. ISSN: 0032-3934.

Publisher: American Chemical Society, Division of Polymer Chemistry.

AB Five different poly 3,4-di((S)-2-methylbutyloxy)alkoxythiophenes contg. increasing lengths of alkoxy spacers were synthesized by the FeCl3 oxidative polymn. of the corresponding monomer. An odd-even effect is obsd. in the optical activity of the .pi.-.pi.* transition of the backbone in the well-aggregated phases formed in n-decanol. The position of the stereocenter in the side chain affects the helicity of the main chain packing. These results are discussed in terms of a cholesteric packing of the polythiophenes.

IT 213691-73-7 213691-76-0 213691-78-2

213691-80-6 213691-83-9

(odd-even effect in assocn. of chiral poly(dialkoxythiophenes))

RN 213691-73-7 HCA

CN Thiophene, 3,4-bis[2-[(2S)-2-methylbutoxy]ethoxy]-, stereoisomer,

homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 213691-72-6 CMF C18 H32 O4 S

Absolute stereochemistry.

RN 213691-76-0 HCA

CN Thiophene, 3,4-bis[3-[(2S)-2-methylbutoxy]propoxy]-, stereoisomer, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 213691-75-9 CMF C20 H36 O4 S

Absolute stereochemistry.

Et
$$S$$
 O $(CH2) 3 O O $(CH2) 3 O S Et $Me$$$

RN 213691-78-2 HCA

CN Thiophene, 3,4-bis[4-[(2S)-2-methylbutoxy]butoxy]-, stereoisomer, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 213691-77-1 CMF C22 H40 O4 S

Absolute stereochemistry.

Et
$$S$$
 O $(CH_2)_4$ O $(CH_2)_4$ O S Et Me

RN 213691-80-6 HCA

CN Thiophene, 3,4-bis[[5-[(2S)-2-methylbutoxy]pentyl]oxy]-, stereoisomer, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 213691-79-3 CMF C24 H44 O4 S

Absolute stereochemistry.

Et
$$S$$
 O $(CH_2)_5$ O $(CH_2)_5$ O S Et Me

RN 213691-83-9 HCA

CN Thiophene, 3,4-bis[[7-[(2S)-2-methylbutoxy]heptyl]oxy]-, stereoisomer, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 213691-82-8 CMF C28 H52 O4 S

Absolute stereochemistry.

Et
$$S$$
 O (CH_2) 7 O S Et Me

CC 36-6 (Physical Properties of Synthetic High Polymers) IT 213691-73-7 213691-76-0 213691-78-2 213691-80-6 213691-83-9 (odd-even effect in assocn. of chiral poly(dialkoxythiophenes)) => d 132 1-29 cbib abs hitstr hitind ANSWER 1 OF 29 HCA COPYRIGHT 2003 ACS on STN 139:76218 Enhanced contrast ratios and rapid-switching color-changeable devices based on poly(3,4-propylenedioxythiophene) derivative and counterelectrode. Xu, Chunye; Tamagawa, Hirohisa; Uchida, Mikio; Taya, Minoru (Univ. of Washington, Seattle, WA, 98195, USA). Proceedings of SPIE-The International Society for Optical Engineering, 4695 (Electroactive Polymer Actuators and Devices (EAPAD)), 442-450 (English) 2002. CODEN: PSISDG. ISSN: 0277-786X. Publisher: SPIE-The International Society for Optical Engineering. A large contrast ratio and rapid switching electrochromic (EC) AB polymer device which consists of laminated two-layer structure between two electrodes was proposed. The new design which only comprises an ITO coated glass electrode, a cathodic poly(3,4-propylenedioxythiophene) deriv. (PProDOT-(CH3)2) EC polymer film, a solid electrolyte and an Au-based counterelectrode which replaces anodic EC polymer and ITO electrode. Carbon-based counterelectrode was prepd. for comparing with Au-based counterelectrode. Lithog. and sputtering were used for Au patterning on glass substrate, while screen printing was used for carbon-based counterelectrode. Covering percentage of Au is less than 20%, in order to keep the electrode high transmittance. authors also prepd. a solid electrolyte, such as poly(Me methacrylate) (PMMA) based contq. LiClO4 gel electrolyte for solid state applications. A special para-film was utilized on sealing the assembly device. Color change of high contrast ratio of transmittance (> .DELTA. 50% T) of the device is rapidly (0.5-1 s) obtained upon applied 2.5 V voltage and repeatable (10,000 times). The temp. range under which the switching is stable is wide, -40.degree. .apprx. 100.degree. C. The repeatability of current of EC polymer devices while color change was estd. by electrochem. 470676-58-5P, Poly(3,3-dimethyl-3,4-dihydro-2H-thieno[3,4-IT b] [1,4] -dioxepine) (cathodic electrochromic layer; large contrast ratio and rapid switching electrochromic polymeric device based on poly(propylenedioxythiophene) deriv. and counterelectrode) 470676-58-5 HCA RN

Poly(3,4-dihydro-3,3-dimethyl-2H-thieno[3,4-b][1,4]dioxepin-6,8-

diyl) (9CI) (CA INDEX NAME)

CN

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes) Section cross-reference(s): 73

electrochromic polymer display device ST

470676-58-5P, Poly(3,3-dimethyl-3,4-dihydro-2H-thieno[3,4-IT b] [1,4] -dioxepine)

(cathodic electrochromic layer; large contrast ratio and rapid switching electrochromic polymeric device based on poly(propylenedioxythiophene) deriv. and counterelectrode)

ANSWER 2 OF 29 HCA COPYRIGHT 2003 ACS on STN L32

139:28503 Electroluminescent element. Gotou, Mariko (Japan). U.S. Pat. Appl. Publ. US 2003113581 A1 20030619, 10 pp. (English). APPLICATION: US 2002-320047 20021216. PRIORITY: JP CODEN: USXXCO. 2001-381941 20011214.

An electroluminescent element is described comprising a AB substrate; a 1st electrode layer formed on a surface of the substrate; an org. electroluminescent layer formed on the 1st electrode layer, the org. electroluminescent layer including at least a luminescent layer; a 2nd electrode layer formed such that the org. electroluminescent layer is interposed between the 1st electrode layer and the 2nd electrode layer; and a sealing base material for sealing the 1st electrode layer, the org. electroluminescent layer and the 2nd electrode layer, wherein the sealing base material is a flexible film and a lamination impact alleviating layer of which universal hardness value is no smaller than 110 N/mm 2 is formed on the 2nd electrode layer. When a flexible film is used as the sealing base material and the flexible film is laminated with the substrate at a predetd. pressure, the org. electroluminescent layer may be prevented from being scratched or cracked. A method of fabricating the

electroluminescent element is also described.

IT 126213-51-2, PEDOT

> (hole transporting material; electroluminescent element having flexible film as sealing base)

126213-51-2 HCA RN

Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX CN

NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

IC ICM H01J001-62

NCL 428690000; 313504000

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 38, 74, 76

ST **electroluminescent** element flexible film impact alleviating layer fabrication

IT Electroluminescent devices

(displays; electroluminescent element having flexible film as sealing base)

IT Electroluminescent devices

Electronic device fabrication

(electroluminescent element having flexible film as sealing base)

IT Luminescent screens

(electroluminescent; electroluminescent

element having flexible film as sealing base)

IT Metals, uses

Oxides (inorganic), uses

Polymers, uses

(impact alleviating layer; **electroluminescent** element having flexible film as sealing base)

IT Polyesters, uses

(substrate, sealing base material; electroluminescent element having flexible film as sealing base)

IT 7440-22-4, Silver, uses 7440-70-2, Calcium, uses 50926-11-9, Indium tin oxide

(electrode; electroluminescent element having flexible film as sealing base)

IT 7631-86-9, Silica, uses

(film on electrode; **electroluminescent** element having flexible film as sealing base)

IT 50851-57-5 **126213-51-2**, PEDOT

(hole transporting material; electroluminescent element having flexible film as sealing base)

IT 25067-59-8, Polyvinyl carbazole

(luminescent layer; electroluminescent element having flexible film as sealing base)

IT 38215-36-0, Coumarin 6

(luminescent layer; electroluminescent element having flexible film as sealing base)

IT 25038-59-9, Polyethylene terephthalate, uses (substrate, sealing base material; electroluminescent element having flexible film as sealing base)

L32 ANSWER 3 OF 29 HCA COPYRIGHT 2003 ACS on STN

138:409984 Self-Aligned, Vertical-Channel, Polymer Field-Effect Transistors. Stutzmann, Natalie; Friend, Richard H.; Sirringhaus, Henning (Cavendish Laboratory, University of Cambridge, Cambridge, CB3 OHE, UK). Science (Washington, DC, United States), 299(5614), 1881-1885 (English) 2003. CODEN: SCIEAS. ISSN: 0036-8075. Publisher: American Association for the Advancement of Science.

The manuf. of high-performance, conjugated polymer transistor AB circuits on flexible plastic substrates requires patterning techniques that are capable of defining crit. features with submicrometer resoln. We used solid-state embossing to produce polymer field-effect transistors with submicrometer crit. features in planar and vertical configurations. Embossing is used for the controlled microcutting of vertical sidewalls into polymer multilayer structures without smearing. Vertical-channel polymer field-effect transistors on flexible poly(ethylene terephthalate) substrates were fabricated, in which the crit. channel length of 0.7 to 0.9 .mu.m was defined by the thickness of a spin-coated insulator layer. Gate electrodes were self-aligned to minimize overlap capacitance by inkjet printing that used the embossed grooves to define a surface-energy pattern.

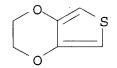
IT 126213-51-2, Poly(3,4-ethylenedioxythiophene)
(PEDOT/PSS; self-aligned, vertical-channel, polymer field-effect transistors)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



CC 76-3 (Électric Phenomena)
Section cross-reference(s): 36

IT 50851-57-5, Poly(styrenesulfonic acid) 126213-51-2, Poly(3,4-ethylenedioxythiophene)

(PEDOT/PSS; self-aligned, vertical-channel, polymer field-effect transistors)

L32 ANSWER 4 OF 29 HCA COPYRIGHT 2003 ACS on STN

138:392673 Red-green-blue light-emitting diodes containing fluorene-based copolymers. Drolet, Nicolas; Beaupre, Serge; Morin, Jean-Francois; Tao, Ye; Leclerc, Mario (Canada Research Chair in Polymer Chemistry, Departement de Chimie, Centre de Recherche en Sciences et Inge, Universite Laval, Quebec City, QC, G1K 7P4, Can.). Journal of Optics A: Pure and Applied Optics, 4(6), S252-S257 (English) 2002. CODEN: JOAOF8. ISSN: 1464-4258. Publisher: Institute of Physics Publishing.

This paper reports the fabrication and evaluation of light-emitting diodes using polyfluorene derivs. as emitter, which cover the entire visible spectral range. Depending on the compn. of the copolymers, red (emission peak at 656 nm), green (488 nm) and blue (428 nm) emission was obtained without any excimer formation. The optimization of the device performances was realized using a multilayered configuration which involves a spin-coated poly(ethylenedioxythiophene) doped with poly(styrene sulfonic acid) (PEDT-PSS) thin film on the ITO anode and an ultrathin LiF layer next to the Al cathode. These 2 layers improve the efficiency of the charge injection. Combining this device configuration with some addnl. charge-transporting mols., luminance at 50-300 cd m-2 have been obtained.

IT 155090-83-8, Baytron P-CH 8000

(Baytron P-CH 8000; red-green-blue light-emitting diodes contg. fluorene-based copolymers)

RN 155090-83-8 HCA

CN Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-51-2 CMF (C6 H6 O2 S)x

CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S

```
CRN 50851-57-5
CMF (C8 H8 O3 S)x
CCI PMS

CM 4

CRN 26914-43-2
CMF C8 H8 O3 S
CCI IDS
```



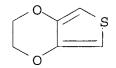
 $D1-CH=CH_2$

D1-SO3H

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 36

IT 155090-83-8, Baytron P-CH 8000 - (Baytron P-CH 8000 - red-green-blue

(Baytron P-CH 8000; red-green-blue light-emitting diodes contg. fluorene-based copolymers)

IT 126213-51-2, Poly(ethylenedioxythiophene) 439588-85-9 527693-03-4 527693-04-5 (red-green-blue light-emitting diodes contg. fluorene-based

copolymers)

L32 ANSWER 5 OF 29 HCA COPYRIGHT 2003 ACS on STN

138:290352 Characterization of polymer solar cells by TOF-SIMS depth profiling. Bulle-Lieuwma, C. W. T.; van Gennip, W. J. H.; van Duren, J. K. J.; Jonkheijm, P.; Janssen, R. A. J.; Niemantsverdriet, J. W. (Philips CFT, Eindhoven, 5656 AA, Neth.). Applied Surface Science, 203-204, 547-550 (English) 2003. CODEN: ASUSEE. ISSN: 0169-4332. Publisher: Elsevier Science B.V..

AB Solar cells consisting of polymer layers sandwiched between a transparent electrode on glass and a metal top electrode are studied using dynamic time-of-flight secondary ion mass spectrometry (TOF-SIMS) in dual-beam mode. Because depth profiling of polymers and polymer-metal stacks is a relatively new field the craters were thoroughly studied by environmental SEM (ESEM), interferometry, surface profilometry and tapping mode AFM. A huge increase in crater bottom roughness was obsd. when starting from the aluminum top layer going in depth, resulting in a loss of depth resoln. layer-to-layer diffusion and contaminants at buried interfaces can be extd. from the depth profiles when taking into account the loss of depth resoln.

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology) Section cross-reference(s): 73, 76

ST polymer heterojunction solar cell TOF SIMS depth profiling multilayer; PEDOT PCBM PPV PSS MDMO polymer layer interface surface

IT Heterojunction solar cells

Interface roughness

Multilayers

Sputtering

Surface structure

TOF-SIMS (time-of-flight secondary-ion mass spectrometry) (characterization of polymer solar cells by TOF-SIMS depth

profiling)

IT Films

(multilayer; characterization of polymer solar cells by TOF-SIMS depth profiling)

IT 177716-59-5, Poly(2-methoxy-5-(3',7'-dimethyl-octyloxy))-p-phenylene vinylene

(MDMO-PPV, neat and blends with PCBM, spin-coated films; characterization of polymer solar cells by

TOF-SIMS depth profiling)

IT 160848-22-6

(PCBM, neat and blends with MDMO-PPV, **spin-coated** films; characterization of polymer solar cells by TOF-SIMS depth profiling)

IT 50851-57-5

(PSS, blend with PEDOT, **spin-coated** films; characterization of polymer solar cells by TOF-SIMS depth profiling)

L32 ANSWER 6 OF 29 HCA COPYRIGHT 2003 ACS on STN

138:228961 Manufacture of **electroluminescent** devices.

Tachikawa, Tomoyuki (Dainippon Printing Co., Ltd., Japan). Jpn.

Kokai Tokkyo Koho JP 2003077652 A2 20030314, 11 pp. (Japanese).

CODEN: JKXXAF. APPLICATION: JP 2001-263372 20010831.

AB The manufg. process comprises the steps of: on a substrate, forming (1) a 1st electrode stripe array; forming (2) a partition wall array perpendicular to (1); forming an org. electroluminescent laminate in (2); and forming a 2nd electrode stripe array perpendicular to (1).

IT 126213-51-2

(prodn. method of electro- luminescent
elements)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

IC ICM H05B033-10 ICS H05B033-14; H05B033-22

- CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
- ST manuf electroluminescent device
- IT Phenolic resins, uses

(novolak; prodn. method of electro- luminescent elements)

IT Coating materials

Electrodes

Electroluminescent devices

Laminated materials

Luminescent substances

Partition

Vapor deposition process

(prodn. method of electro-luminescent

elements)

IT Polyimides, uses

Polysiloxanes, uses

(prodn. method of electro- luminescent

elements)

IT 1317-70-0, Anatase 7440-22-4, Silver, uses 7440-70-2, Calcium, uses 37221-36-6, PSS 50926-11-9, ITO 95270-88-5, Polyfluorene 126213-51-2

(prodn. method of electro- luminescent
elements)

- L32 ANSWER 7 OF 29 HCA COPYRIGHT 2003 ACS on STN
- 138:160835 Organic electroluminescent devices containing a red emissive layer contains blue, green and red emissive materials and a method for manufacturing the devices. Tada, Takumu (Japan). U.S. Pat. Appl. Publ. US 2003030370 A1 20030213, 23 pp. (English). CODEN: USXXCO. APPLICATION: US 2002-214259 20020808. PRIORITY: JP 2001-245734 20010813; JP 2002-123144 20020424.
- Org. electroluminescent devices are described which AB comprise a plurality of first electrodes disposed on a substrate in matrix; a second electrode disposed with being confronted with each of the plurality of first electrodes; and a emissive layer formed between each of the plurality of first electrodes and the second electrode on each of the plurality of first electrodes, where the emissive layer is composed of a blue (B) emissive layer, a green (G) emissive layer and a red (R) emissive layer as a set of pixels, the org. electroluminescence element is further characterized in that the B emissive layer contains a B emissive material, the G emissive layer contains B and G emissive materials, and the R emissive layer contains B, G and R emissive materials. A method for manufg. the org. electroluminescent device is discussed which entails forming a plurality of first electrodes divided by a plurality of separators and disposed on a substrate in matrix; forming a B emissive layer on the plurality of first electrodes by diffusing a B emissive material; obtaining a G emissive layer adjacent to the B emissive layer after diffusing a G emissive material in a part of the B emissive layer; obtaining a R emissive layer adjacent to the G emissive layer after diffusing a R

emissive material in a part of the G emissive layer; and forming a second electrode on each of the R and G and B emissive layers.

IT 126213-51-2, Polyethylene dioxythiophene

(layer contg. polystyrenesulfonate and; org.

electroluminescent devices contg. red emissive layer
contains blue, green and red emissive materials and method for

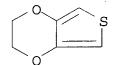
manufg. devices)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



IC ICM H05B033-14

NCL 313504000

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 22, 76

ST org electroluminescent device fabrication diffusion; OLED blue green red emissive layer diffusion

IT Diffusion

IT

(emissive layer formed by; org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

IT Electroluminescent devices

Electronic device fabrication .

(org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

emissive materials and method for manufg. devices)
7440-22-4, Silver, uses 7440-70-2, Calcium, uses
(cathode layer; org. electroluminescent devices contg.

red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

IT 38215-36-0, Coumarin-6

(film on Si substrate, green-emitting material; org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

IT 50926-11-9, Indium tin oxide

(glass substrate coated with; org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

IT 50851-57-5

(layer contg. polyethylene dioxythiophene and; org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

- IT 15082-28-7, 2-(4-Biphenylyl)-5-(4-tert-butylphenyl)-1,3,4-oxadiazole 25067-59-8, Poly(N-vinylcarbazole)

(org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

IT 51325-95-2

(red-emitting material; org. **electroluminescent** devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

IT 7440-57-5, Gold, uses

(release layer; org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

IT 7440-21-3, Silicon, uses

(substrate; org. electroluminescent devices contg. red emissive layer contains blue, green and red emissive materials and method for manufg. devices)

- L32 ANSWER 8 OF 29 HCA COPYRIGHT 2003 ACS on STN .
- 138:128647 Fully transparent, organic light-emitting electrochemical cells. Ouisse, T.; Armand, M.; Kervella, Y.; Stephan, O. (Laboratoire de Spectrometrie Physique, Universite Joseph Fourier Grenoble 1 and CNRS, Saint-Martin d'He`res, 38042, Fr.). Applied Physics Letters, 81(17), 3131-3133 (English) 2002. CODEN: APPLAB. ISSN: 0003-6951. Publisher: American Institute of Physics.
- The authors report the fabrication and performance of fully transparent, org. blue light-emitting electrochem. cells (OLECs), in which both the anode and cathode are made of In Sn oxide. The active layer is a blend of polyfluorene with long and flexible alkyl side chains grafted on the 9,9 position and of a molten salt. Two identical spin-coated active layers are

laminated together at high temp. to form the OLECs. The electroluminescence threshold is .apprx.3.3 V and the light intensity exceeds 10 .mu.W/cm2 at 5 V.

IT 155090-83-8

(fully transparent, org. light-emitting electrochem. cells contq.)

RN 155090-83-8 HCA

CN Benzenesulfonic acid, ethenyl-, homopolymer, compd. with

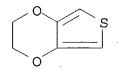
2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-51-2 CMF (C6 H6 O2 S)x CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S



CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



 $D1-CH=CH_2$

 $D1-SO_3H$

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
Section cross-reference(s): 36

IT Electric current-potential relationship Electron-hole recombination

Lamination

Luminescence

IT

Luminescence, electroluminescence

Radiative recombination

(fully transparent, org. light-emitting electrochem. cells with) 50926-11-9, ITO 155090-83-8 268536-01-2,

Tetrahexylammonium-bis(trifluoromethylsulfonyl)imide 268536-02-3 (fully transparent, org. light-emitting electrochem. cells contg.)

L32 ANSWER 9 OF 29 HCA COPYRIGHT 2003 ACS on STN

138:114821 Organic light-emitting elements which can employ non-volatile or insoluble materials and light-emitting devices using the elements. Seo, Satoshi; Murakami, Masakazu; Yamazaki, Shunpei (Semiconductor Energy Laboratory Co., Ltd., Japan). U.S. Pat. Appl. Publ. US 2003015960 A1 20030123, 36 pp. (English). CODEN: USXXCO. APPLICATION: US 2002-158233 20020531. PRIORITY: JP 2001-167508 20010601; JP 2001-167662 20010604.

Org. light-emitting elements are described which AB comprise an org. compd. layer sandwiched between an anode and a cathode, the org. compd. layer comprising a mixed layer having a plurality of org. compds. serving as a host material and a guest material, where .gtoreq.1 of the org. compds. that serve as the host material forms a uniform amorphous film, and where the quest material forms an aggregation having a diam. .ltoreg.1 than the thickness of the mixed layer. emitting devices are described having a lightemitting element comprising an anode; a cathode, and an org. compd. layer placed between the anode and the cathode, the org. compd. layer contg. plural kinds of org. compds., where the org. compd. layer comprises a plurality of orq. compds. that form a uniform amorphous film and .gtoreq.1 of org. compds. that are scattered in the amorphous film in the form of granule having a diam. < the thickness of the org. compd. layer.

IT 126213-51-2, Poly(ethylene dioxythiophene)

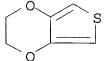
(hole injection layer contg.; org. lightemitting elements which can employ non-volatile or insol. materials and light-emitting devices using elements)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



```
IC
     ICM H05B033-00
     313504000
NCL
CC
     73-11 (Optical, Electron, and Mass Spectroscopy and Other Related
     Properties)
     Section cross-reference(s): 74, 76
ST
     org light emitting element device granule
     aggregate material
     Electroluminescent devices
IT
         (displays; org. light-emitting
        elements which can employ non-volatile or insol. materials and
        light-emitting devices using elements)
     Luminescent screens
IT
         (electroluminescent; org. light-
        emitting elements which can employ non-volatile or insol.
        materials and light-emitting devices using
        elements)
IT
     Fluorescent substances
     Phosphorescent substances
         (granules or guest material forming aggregates; org.
        light-emitting elements which can employ
        non-volatile or insol. materials and light-
        emitting devices using elements)
     Electric appliances
ΙT
       Electroluminescent devices
         (org. light-emitting elements which can
        employ non-volatile or insol. materials and light-
        emitting devices using elements)
IT
     Aggregates
     Granular materials
         (orq. light-emitting elements which can
        employ non-volatile or insol. materials in form of)
     7440-21-3, Silicon, uses
IT
         (cryst. film; org. light-emitting elements
        which can employ non-volatile or insol. materials and
        light-emitting devices using elements)
IT
     7631-86-9, Silica, uses
         (gate insulating film; org. light-emitting
        elements which can employ non-volatile or insol. materials and
        light-emitting devices using elements)
     50851-57-5, Poly (styrenesulfonic acid) 126213-51-2,
IT
     Poly(ethylene dioxythiophene)
         (hole injection layer contq.; org. light-
        emitting elements which can employ non-volatile or insol.
        materials and light-emitting devices using
        elements)
```

- IT 15082-28-7, 2-(4-Biphenyl)-5-(4-tert-butylphenyl)-1,3,4-oxadiazole 25067-59-8, Poly(N-vinylcarbazole) (host mixt. contg.; org. light-emitting

elements which can employ non-volatile or insol. materials and light-emitting devices using elements)

- L32 ANSWER 10 OF 29 HCA COPYRIGHT 2003 ACS on STN

 138:98008 Light-emitting devices and methods of
 manufacturing the devices involving simplified formation of a
 laminate structure of organic films deposited from solutions
 in protic and aprotic solvents. Ogino, Kiyofumi; Shibata, Noriko
 (Semiconductor Energy Laboratory Co., Ltd., Japan). U.S. Pat. Appl.
 Publ. US 2003006699 A1 20030109, 26 pp. (English). CODEN: USXXCO.
 APPLICATION: US 2002-177752 20020624. PRIORITY: JP 2001-191678
 20010625.
- Methods of manufq. light-emitting devices are ABdiscussed which entail forming a 2nd org. compd. layer on a 1st org. compd. layer; forming a 1st conductive film on the 2nd org. compd. layer; etching a portion of the 2nd org. compd. layer by a wet etching, where the portion of the 2nd org. compd. layer does not overlap the 1st conductive film; forming a 3rd org. compd. layer on the 1st org. compd. layer; forming a 2nd conductive film on the 3rd org. compd. layer; etching a portion of the 1st org. compd. layer by a dry etching, where the portion of the 1st org. compd. layer does not overlap the 1st and 2nd conductive films, where the 1st org. compd. layer is formed by applying a soln. including a protic solvent, where each of the 2nd and 3rd org. compd. layers is formed by applying a soln. including an aprotic solvent. Lightemitting devices are described which comprise a 1st light-emitting element including a 1st anode; a 1st org. compd. layer in contact with the 1st anode; a 2nd org. compd. layer in contact with the 1st org. compd. layer; a 1st cathode in contact with the 2nd org. compd. layer; a 2nd light-emitting element including a 2nd anode; a 3rd org. compd. layer in contact with the 2nd anode; a 4rt org. compd. layer in contact with the 3rd org. compd. layer; a 2nd cathode in contact with the 4th org. compd. layer; a 3rd light-emitting element including a 3rd anode; a 5th org. compd. layer in contact with the 3rd anode; a 6th org. compd. layer in contact with the 5th org. compd. layer; a 3rd

cathode in contact with the 6th org. compd. layer; a conductive film in contact with the 1st, 2nd and 3rd cathodes. ΙT 126213-51-2, PEDOT (light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) 126213-51-2 HCA RNThieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX CNNAME) CM CRN 126213-50-1 CMF 'C6 H6 O2 S ICM H05B033-00 IC 313506000 NCL 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties) Section cross-reference(s): 38, 74, 76 ST electroluminescent device fabrication org film laminate protic aprotic solvent; OLED manufg film deposition protic aprotic solvent Solvents IT (aprotic; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) IT Polyacetylenes, uses (deriv.; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) Electroluminescent devices IT (displays; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) ΙT Etching (dry; light-emitting devices and methods of .

manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in

IT Luminescent screens (electroluminescent; light-emitting

protic and aprotic solvents and use of)

devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) Electroluminescent devices IT Electronic device fabrication (light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) IT Etching (light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents and use of) ΙT Etching (plasma, oxygen; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents and use of) IT Solvents (protic; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) IT 50926-11-9, Indium tin oxide (anode; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) 67-68-5, Dimethyl sulfoxide, uses IT67-66-3, Chloroform, uses 71-43-2, Benzene, uses 75-09-2, Dichloromethane, uses 96-48-0, .gamma.-Butyrolactone 100-66-3, Anisole, uses 108-88-3, Toluene, 108-90-7, Chlorobenzene, uses 108-94-1, Cyclohexanone, uses 109-99-9, Tetrahydrofuran, uses 110-82-7, Cyclohexane, uses 111-76-2, Butylcellosolve 119-64-2, Tetralin 123-91-1, Dioxane, 1330-20-7, Xylene, 872-50-4, N-Methyl-2-pyrrolidone, uses uses 25321-22-6, Dichlorobenzene (aprotic solvent, etchant; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) 7429-90-5, Aluminum, uses IT(auxiliary electrode; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) 95270-88-5D, Polyfluorene, dialkyl deriv. IT(blue-emitting layer; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited

from solns. in protic and aprotic solvents)

12798-95-7

IT

(cathodes; light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents)

- IT 26009-24-5D, Poly1,4-phenylene vinylene, dialkoxyphenyl derivs.

 (green-emitting layer; light-emitting
 devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents)
- 9033-83-4D, Polyphenylene, alkyl derivs. 25067-58-7D, IT Polyacetylene, deriv. 25190-62-9D, Poly(1,4-phenylene), deriv. 25190-62-9D, Poly(1,4-phenylene), dialkoxy deriv. 25233-30-1, 25233-34-5D, Polythiophene, alkyl deriv. 25233-34-5D, Polythiophene, deriv. 26009-24-5D, Poly(1,4-phenylene-1,2-ethenediyl), deriv. Polyfluorene, deriv. 98705-03-4, Polyhexylphenylacetylene 120659-35-0, 104934-50-1, Poly(3-hexylthiophene) Poly(3-cyclohexylthiophene) 126213-51-2, PEDOT 141807-85-4, Poly[3-(4-octylphenyl)thiophene] 138184-36-8 157673-32-0 159838-09-2, Poly[3-(4-octylphenyl)-2,2'-bithiophene] 163045-79-2, Poly(3-cyclohexyl-4-methylthiophene) 195456-48-5, Poly(9,9-dioctyl-9H-fluorene-2,7-diyl) 220613-28-5 482373-10-4 (light-emitting devices and methods of manufg. the devices involving simplified formation of laminate structure of org. films deposited from solns. in protic and aprotic solvents) .
- L32 ANSWER 11 OF 29 HCA COPYRIGHT 2003 ACS on STN

 137:325878 Multi-layer polymer light-emitting diodes
 with 2,3-dialkoxy-p-phenylene vinylene and its blends. Sano,
 Takeshi; Tuan, Chi-Shen; Martin, Rainer E.; Holmes, Andrew B.
 (Materials and Devices Development Center, SANYO Electric Co., Ltd.,
 Osaka, 573-8534, Japan). Journal of Photopolymer Science and
 Technology, 15(2), 253-258 (English) 2002. CODEN: JSTEEW. ISSN:
 0914-9244. Publisher: Technical Association of Photopolymers,
 Japan.
- AB A green-fluorescent polymer, poly(2,3-dibutoxy-1,4-phenylene vinylene) (DB-PPV) was synthesized via dehydro-halogenation polymn. of 2,3-dibutoxy-1,4-bis(bromomethyl)benzene using K tert-butoxide initiator in dry THF, to obtain DB-PPV as yellow fibers. The photoluminescence (PL) peak wavelength of DB-PPV in soln. is 492 nm and the PL quantum yield in chloroform is 72%; the PL peak wavelength of spin-coated films is 522 nm and PL quantum yield is 22%. Electroluminescent PLED devices were assembled using various layers of DB-PPV; poly(3,4-ethylenedioxithiophene): poly(styrenesulfonic acid) (PEDOT:PSS) as buffer layer and hole transport layer; and 1,3-bis[5-(p-t-butyl-

phenyl)-1,3,,4-oxadiazol-2-yl]benzene (OXD-7) and tris(8-hydroxy quinolinato)aluminum (Alq3) as electron transport-layer. The EL efficiency of ITO/DB-PPV/Ca/Al devices improved when an electron transport-layer was incorporated into the structure. A blend of DB-PPV and poly(9,9-dioctylfluorene) (PF8) was also used in PLED structures; the EL peak wavelength was blue-shifted to 503 nm and the EL efficiency improved.

IT 126213-51-2, Poly(3,4-ethylenedioxythiophene)

(buffer and hole transport-layer; electroluminescence efficiency of multi-layer PLEDS with prepd.

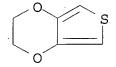
poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



CC 36-5 (Physical Properties of Synthetic High Polymers)

Section cross-reference(s): 35, 73

IT Polymerization

(dehydro-halogenation; electroluminescence efficiency of multi-layer PLEDS with prepd.

poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter)

IT Fluorescence

Hole transport

Luminescence

(electroluminescence efficiency of multi-layer

PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter)

IT Polymer blends

(electroluminescence efficiency of multi-layer

PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter)

IT Poly(arylenealkenylenes)

(electroluminescence efficiency of multi-layer

PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter)

IT Electroluminescent devices

(green-emitting; electroluminescence efficiency of multi-layer PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter)

- 50851-57-5, Poly(styrenesulfonic acid) 126213-51-2, IT Poly(3,4-ethylenedioxythiophene) (buffer and hole transport-layer; electroluminescence efficiency of multi-layer PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter) 7440-70-2, Calcium, uses IT 7429-90-5, Aluminum, uses (cathodes; electroluminescence efficiency of multilayer PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter) 50926-11-9, ITO IT (contact layer; electroluminescence efficiency of multi -layer PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene) vinylene) and blend with polyfluorene emitter) IT 208264-13-5P 224456-13-7P (electroluminescence efficiency of multi-layer PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter) 138372-67-5, 1,3-Bis[5-(p-tert-butyl-phenyl)-2085-33-8, Alq3 IT 1,3,,4-oxadiazol-2-yl]benzene (electron transport-layer; electroluminescence efficiency of multi-layer PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter) 195456-48-5, Poly(9,9-dioctyl-9H-fluorene-2,7-diyl) IT (poly(9,9-dioctylfluorene); electroluminescence efficiency of multi-layer PLEDS with prepd. poly(2,3-dibutoxy-p-phenylene vinylene) and blend with polyfluorene emitter) ANSWER 12 OF 29 HCA COPYRIGHT 2003 ACS on STN 137:317570 Novel heterolayer organic light-emitting diodes based on a conjugated dendrimer. Ma, Dongge; Lupton, John M.; Beavington, Richard; Burn, Paul L.; Samuel, Ifor D. W. (School of Physics and Astronomy, University of St. Andrews, St. Andrews, KY16 9SS, UK). Advanced Functional Materials, 12(8), 507-511 (English) 2002. CODEN: AFMDC6. ISSN: 1616-301X. Publisher: Wiley-VCH Verlag GmbH. We demonstrate a novel org. light-emitting diode AΒ (LED) heterolayer structure that contains a conjugated dendrimer as the light-emitting mol. The LED was prepd. by spin-coating two dendrimer layers from the same The device consists of a graded bilayer structure formed solvent. from a neat dendrimer film covered with a film consisting of the same dendrimer but doped with the electron-transporting material 2-(4-biphenylyl)-5-phenyl-1,3,4-oxadiazole (PBD). In this device, the heterojunction interface present in conventional bilayer org. light-emitting diodes is eliminated, and is replaced by a graded interlayer. By optimizing the concn. of PBD in
 - external quantum efficiency of 0.16% at 600 cd m-2 was obtained. The **EL** quantum efficiency is significantly enhanced in

the dendrimer, a peak electroluminescence (EL)

comparison with devices based on a single layer, a conventional bilayer, and a single-layer doped with PBD. The **EL** quantum efficiency is a factor of 8 larger than that of a conventional bilayer LED made with the conjugated dendrimer as the emissive layer and poly(methylmethacrylate) (PMMA) doped with PBD as the electron-transporting layer. The best blended device exhibited only one third of the efficiency of the graded device. The improvement in the operating characteristics of the graded device is attributed to the efficient device structure, in which exciton formation is improved by a graded doping profile of electron- and hole-transporting components.

IT 155090-83-8

(novel heterolayer org. light-emitting diodes based on conjugated dendrimer deposited sequentially to form graded multilayer device contg.)

RN 155090-83-8 HCA

CN Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-51-2 CMF (C6 H6 O2 S)x CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S

CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



 $D1-CH=CH_2$

 $D1-SO_3H$

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related
Properties)
Section cross-reference(s): 38, 76
ST heterolayer org light emitting diode conjugated
dendrimer electroluminescence; OLED graded
dendrimer PBD doped PEDOT polystyrenesulfonate spin
coating
IT Dendritic polymers
 (conjugated; novel heterolayer org. lightemitting diodes based on conjugated dendrimer deposited
sequentially to form graded multilayer device)

IT **Electroluminescent** devices
(novel heterolayer org. **light-emitting** diodes
based on conjugated dendrimer deposited sequentially to form
graded **multilayer** device)

IT Electric current-potential relationship
Luminescence, electroluminescence
(of heterolayer org. light-emitting diodes
based on conjugated dendrimer)

IT 852-38-0, 2-(4-Biphenylyl)-5-phenyl-1,3,4-oxadiazole (dopant; novel heterolayer org. light-emitting diodes based on conjugated dendrimer deposited sequentially to form graded multilayer device contg.)

One of the state o

IT 7429-90-5, Aluminum, uses 7440-70-2, Calcium, uses 50926-11-9, ITO

(novel heterolayer org. light-emitting diodes based on conjugated dendrimer deposited sequentially to form graded multilayer device contg.)

```
ANSWER 13 OF 29 HCA COPYRIGHT 2003 ACS on STN
L32
137:224259
            Thermal imaging processes and products of electroactive
     organic material. Blanchet-Fincher, Graciela Beatriz (E. I. Du Pont
     de Nemours and Company, USA). PCT Int. Appl. WO 2002070271 A2
     20020912, 33 pp. DESIGNATED STATES: W: AE, AG, AL, AM, AT, AU, AZ,
     BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ,
     EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ,
     TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW, AM, AZ, BY,
     KG, KZ, MD, RU, TJ, TM; RW: AT, BE, BF, BJ, CF, CG, CH, CI, CM, CY,
     DE, DK, ES, FI, FR, GA, GB, GR, IE, IT, LU, MC, ML, MR, NE, NL, PT,
     SE, SN, TD, TG, TR. (English). CODEN: PIXXD2. APPLICATION: WO
     2002-US8164 20020221. PRIORITY: US 2001-PV272440 20010301.
     Processes for effecting thermal transfer of electroactive org.
AB
     material are disclosed wherein unwanted portions of a layer of
     electro org. material supported by a donor element are removed or
     transferred from the layer by thermal transfer, particularly
     laser-induced thermal transfer, leaving a desired pattern of the
     electroactive org. material on the donor element.
                                                           The electroactive
     org. material may be an org. material exhibiting
     electroluminescence, charge transport, charge injection,
     elec. cond., semicond. and/or exciton blocking.
                                                         The layer of
     electroactive org. material may comprise more than one layer of
     different types of electroactive org. material.
                                                        The exposure
     pattern is a neg. image of the desired pattern.
                                                         The electroactive
     org. material of the desired pattern is not, therefore, exposed to
     the heat which can cause decompn. The desired pattern of
     electroactive org. material may then be gently transferred from the
     donor element to a desired substrate by lamination, for
     example, without damaging the electroactive org. material.
     substrate may be used to form orq. electronic devices, such as
     light emitting displays, photodetectors and
     photovoltaic cells. Donor elements for use in the processes are
     also disclosed.
ΙT
     155090-83-8, Baytron P
        (charge injection layer; thermal imaging processes and products
        of electroactive org. material contg.)
     155090-83-8
RN
                 HCA
     Benzenesulfonic acid, ethenyl-, homopolymer, compd. with
CN
     2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX
     NAME)
```

CM 1

CRN 126213-51-2 CMF (C6 H6 O2 S)x CCI PMS CM 2

CRN 126213-50-1 CMF C6 H6 O2 S

CM

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI **PMS**

CM

CRN 26914-43-2 C8 H8 O3 S CMF CCI IDS

 $D1-CH=CH_2$

 $D1-SO_3H$

IC ICM B41M003-00

ICS B41M007-00

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes) Section cross-reference(s): 76

thermal transfer process light emitting ST

device display

Electroluminescent devices IT

Optical detectors

Photoelectric devices

(thermal imaging processes and products of electroactive org. material for)

IT 50851-57-5, Polystyrene sulfonic acid 155090-83-8, Baytron P 427889-73-4, XICP-OS 01 (charge injection layer; thermal imaging processes and products of electroactive org. material contg.)

L32 ANSWER 14 OF 29 HCA COPYRIGHT 2003 ACS on STN

137:224253 Multilayer arrangement with conductive polymer for flat panel electro-optical display. Wehrmann, Rolf; Heuer, Helmut-Werner; Karbach, Alexander (Bayer Aktiengesellschaft, Germany). Eur. Pat. Appl. EP 1239322 A2 20020911, 7 pp. DESIGNATED STATES: R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR. (German). CODEN: EPXXDW. APPLICATION: EP 2002-3272 20020222. PRIORITY: DE 2001-10110755 20010307; DE 2001-10127401 20010606.

Ι

The invention relates to a **multilayer** arrangement including an elec. conductive polymer layer-contg: transparent substrate applicable for a flat **panel** electro-optical **display** like a liq. crystal display, electrochromic display, etc., wherein the conductive polymer is polyaniline, polypyrrol, or polythiophene represented by I (A1, A2 = C1-18-alkyl, C1-18-alkylene; n = 2-10,000). The transparent substrate may comprise a glass or polymer. The conductive polymer layer reduced the surface roughness of the substrate.

RN 155090-83-8 HCA

CN Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-51-2 CMF (C6 H6 O2 S)x CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S

CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



 $D1-CH=CH_2$

D1-S03H

IC ICM G02F001-1343 ICS G02F001-1333; H01B001-12; C08G061-12; C09D005-24

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 38

ST multilayer substrate conductive polymer flat panel electrooptical display polythiophene

IT Polyanilines

(conductive polymer in multilayer substrate for flat panel electro-optical display)

IT Conducting polymers
Electrochromic imaging devices

Electrooptical imaging devices

Liquid crystal displays

(multilayer substrate with conductive polymer for flat panel electro-optical display)

IT Conducting polymers

(polypyrroles; conductive polymer in multilayer substrate for flat panel electro-optical display)

ITConducting polymers

(polythiophenes; conductive polymer in multilayer substrate for flat panel electro-optical display)

IT Glass, uses

Polyesters, uses

Polyesters, uses

(transparent substrate; multilayer substrate with conductive polymer for flat panel electro-optical display)

155090-83-8, Baytron P IT

> (conductive polymer in multilayer substrate for flat panel electro-optical display)

25038-59-9, Polyethyleneterephthalate, uses IT (transparent substrate; multilayer substrate with conductive polymer for flat panel electro-optical display)

L32 ANSWER 15 OF 29 HCA COPYRIGHT 2003 ACS on STN

137:12919 Sol-gel-deposited Sb-doped SnO2 as transparent anode for **OLED**: process, patterning, and hole injection characteristics. Vaufrey, David; Ben Khelifa, M.; Besland, Marie-Paule; Sandu, C.; Blanchin, Marie-Genevieve; Teodorescu, Valentin S.; Roger, Jean-Alain; Tardy, Jacques (Laboratoire d'Electronique, Optoelectronique et Microsystemes, (UMR CNRS 5512), Ecole Centrale de Lyon, Ecully, 69131, Fr.). Proceedings of SPIE-The International Society for Optical Engineering, 4464 (Organic Light-Emitting Materials and Devices V), 103-112 (English) 2002. ISSN: 0277-786X. Publisher: SPIE-The International CODEN: PSISDG. Society for Optical Engineering.

AB This paper reports on the potentialities of sol-gel deposited Sb doped SnO2(TO) as a new transparent conducting oxide (TCO) for anode in orq. light emitting diodes (OLED).

Multilayered films with transparency over 85% and Structural resistivity <5 103.OMEGA. -cm were obtained. observations by TEM show that the films are nanocrystd. Smaller and more uniform grains are obtained upon rapid thermal annealing. At. Force Microscopy (AFM) imaging shows the surface roughness does not TO films are very stable and cannot be chem. etched. exceed 20 A. Anode patterning by reactive ion etching (RIE) in a Methane- H plasma was experienced and is described. Typical etching around 250 TO/PEDOT/PVK/Al hole only diodes were realized A/min were obtained. to assess sol gel TO films as hole injection electrodes. with threshold voltages of 6 V were obtained. A comparison with ITO deposited by low temp. cathodic sputtering is given.

IT 126213-51-2, PEDOT

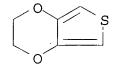
(antimony-doped tin dioxide sol-gel-deposited transparent anode for **LED** contg.)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT Sol-gel processing

(-deposited antimony-doped tin dioxide as transparent anode for LED)

IT Atomic force microscopy

Electric current-potential relationship

Electric resistance

Surface roughness

Transmission electron microscopy

(of sol-gel-deposited antimony-doped tin dioxide as transparent anode for **LED**)

IT Transparent materials

(sol-gel-deposited antimony-doped tin dioxide as LED anode)

IT Anodes

(sol-gel-deposited antimony-doped tin dioxide as **LED** transparent)

IT **Electroluminescent** devices

(sol-gel-deposited antimony-doped tin dioxide as transparent anode for)

IT 7440-36-0, Antimony, uses

(-doped tin dioxide sol-gel-deposited transparent anode for LED)

IT 18282-10-5, Tin dioxide

(antimony-doped sol-gel-deposited transparent anode for **LED**)

IT 25067-59-8, 9H-Carbazole, 9-ethenyl-, homopolymer

126213-51-2, PEDOT

(antimony-doped tin dioxide sol-gel-deposited transparent anode for **LED** contg.)

L32 ANSWER 16 OF 29 HCA COPYRIGHT 2003 ACS on STN

137:12916 High-performance flexible polymer light-emitting diodes fabricated via a low-temperature plastic laminated process. Guo, Tzung-Fang; Chang, Shun-Chi; Pyo, Seungmoon; Yang, Yang (Department of Materials Science and Engineering, University of California-Los Angeles, Los Angeles, CA, 90095-1595, USA). Proceedings of SPIE-The International Society for Optical Engineering, 4464 (Organic Light-Emitting Materials and Devices V), 34-41 (English) 2002. CODEN: PSISDG. ISSN: 0277-786X. Publisher: SPIE-The International Society for Optical Engineering.

The fabrication of high performance polymer LEDs (PLEDs) using a low temp., plastic lamination process is reported. Blue- and red-emitting PLEDs were fabricated by laminating different luminescent polymers and org. compds. together to form the active media. This unique approach eliminates the issue of org. solvent compatibility with the org. layers for fabricating multi-layer PLEDs. A template activated surface process (TAS) was applied to generate an optimum interface for the low temp. lamination process. The at. force microscopy anal. reveals a distinct difference in the surfaces created by the TAS and the spin-coating process. This observation coupled with the device data confirms the importance of the activated interface in the lamination process.

IT 126213-51-2, PEDOT

(high-performance flexible LED fabricated via low-temp. plastic laminated process)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 38, 76

ST light emitting diode flexible polymer low temp plastic laminated; LED flexible polymer low temp plastic laminated

IT Polymers, uses

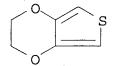
(high-performance flexible LED fabricated via low-temp. plastic laminated process)

IT Laminated plastics, uses

(high-performance flexible polymer LED fabricated via low-temp.

```
process)
IT
     Atomic force microscopy
     Luminescence, electroluminescence
         (of high-performance flexible polymer LED fabricated via
        low-temp. plastic laminated process)
IT
     Electroluminescent devices
        (of high-performance flexible polymer fabricated via low-temp.
        plastic laminated process)
IT
     Coating process
         (spin; of high-performance flexible polymer LED
        fabricated via low-temp. plastic laminated process)
     123847-85-8, .alpha.-NPD
                                123864-00-6, Poly(9,9-dioctylfluorene)
ΙT
     126213-51-2, PEDOT
                           138184-36-8, MEH-PPV
         (high-performance flexible LED fabricated via low-temp. plastic
        laminated process)
     ANSWER 17 OF 29 HCA COPYRIGHT 2003 ACS on STN
           Light emitting device using triplet
136:332912
                Yamazaki, Shunpei; Nishi, Takeshi; Mizukami, Mayumi;
     compound.
     Ikeda, Hisao (Japan). U.S. Pat. Appl. Publ. US 20020050786 Al
     20020502, 43 pp.
                       (English).
                                    CODEN: USXXCO.
                                                     APPLICATION: US
     2001-938291 20010824. PRIORITY: JP 2000-258260 20000828.
    A light emitting device is described comprising
AB
     a substrate having a pixel portion; and a plurality of EL
     elements in the pixel portion, at least one of the EL
     elements comprising an EL layer comprising a triplet
     compd. (e.g., CBP and Ir(ppy)3), wherein the EL
     layer comprises a plurality of hole transporting
     layers contq. MTDATA and layers contq. .alpha.-NPD, and a hole
     injection layer comprising copper phthalocyanine. The luminance of different colors of light emitted from
     EL elements in a pixel portion of a light
     emitting device is equalized and the luminance of
     light emitted from the EL elements is
              A hole transporting layer has a laminate
     raised.
     structure to thereby cause the EL elements to emit
     light of higher luminance. An elec. appliance (e.g, video
     camera, imaging device, recording medium, personal computer, cellular phone, audio reproducing device) having a light
     emitting device is also described comprising a substrate
     having a pixel portion; and a plurality of EL elements in
     the pixel portion, at least one of the EL elements
     comprising an EL layer comprising a triplet compd.,
     wherein the EL layer comprises a
     plurality of hole transporting layers.
     126213-51-2, PEDOT
IT
         (light emitting device using triplet compd.)
RN
     126213-51-2 HCA
     Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX
CN
     NAME)
```

CRN 126213-50-1 CMF C6 H6 O2 S



IT

IC ICM H05B033-14

NCL 313504000

74-13 (Radiation Chemistry, Photochemistry, and Photographic and CC Other Reprographic Processes) Section cross-reference(s): 73, 76

STorq light emitting display imaging device

Electric appliances IT

Electroluminescent devices

Optical imaging devices

(light emitting device using triplet compd.)

IT

7440-44-0, Carbon, processes (diamond-like films; light emitting device using triplet compd.)

7440-64-4, Ytterbium, processes

(film; light emitting device using triplet compd.)

7723-14-0, Phosphorus, processes 7440-56-4, Germanium, processes IT (light emitting device using triplet compd.)

1314-13-2, Zinc oxide (ZnO), IT 147-14-8, Copper phthalocyanine 7440-21-3, Silicon, processes processes 2085-33-8, AlQ3 7631-86-9, Silicon oxide, processes 7440-33-7, Tungsten, processes 11105-01-4, Silicon oxynitride 12024-08-7, Gallium oxide (GaO) 12033-62-4, Tantalum nitride 26009-24-5, Poly(1,4-phenylene-1,2-58328-31-7, 4,4'-Bis(carbazol-9-yl)biphenyl ethenediyl) 94928-86-6, Tris(2-phenylpyridine)iridium 123847-85-8, .alpha.-NPD 124729-98-2, MTDATA 126213-51-2, PEDOT

(light emitting device using triplet compd.)

ANSWER 18 OF 29 HCA COPYRIGHT 2003 ACS on STN

136:238752 Controlling exciton diffusion in multilayer white phosphorescent organic light emitting devices. D'Andrade, Brian W.; Thompson, Mark E.; Forrest, Stephen R. (Center for Photonics and Optoelectronic Materials (POEM), Princeton Materials Institute (PMI), Department of Electrical Engineering, Princeton University, Princeton, NJ, 08544, USA). Advanced Materials (Weinheim, Germany), 14(2), 147-151 (English) 2002. ISSN: 0935-9648. Publisher: Wiley-VCH Verlag GmbH. CODEN: ADVMEW.

ABThe combination of 2 multilayer org. light emitting diodes and blue, yellow, and red phosphor doped emissive regions was used to efficiently produce white light. Two white OLED (WOLED) structures were used, i.e., device 1 is

a 3 phosphor structure and device 2 is a blocking layer structure. At .lambda. = 520-600 nm, device 2 had almost no electroluminescent spectra emission, while device 1 had considerably more emission from bis(2-phenylbenzothiozolato-N-C2)iridium(acetylacetonate) (Bt2Ir(acac)) in this region. The addnl. doped layer improved the efficiency of device 2 as compared to device 1 by boosting the yellow emission where the human eye had the highest photonic response efficiency, and using Bt2Ir(acac). The multi-emissive layer fully electrophosphorescent WOLEDs could take advantage of the diffusion

electrophosphorescent WOLEDs could take advantage of the diffusion of triplets to produce bright white devices with high power and quantum efficiencies. The device color could be tuned by varying the thickness and the dopant concns. in each layer, and by introducing exciton blocking layers between emissive layers.

IT 126213-51-2, Poly(3,4-ethylenedioxythiophene)

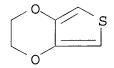
(controlling exciton diffusion in multilayer white phosphorescent org. light emitting devices)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



IT

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 38, 76

ST white polymer light emitting diode

multilayer exciton diffusion Electroluminescent devices

Exciton

Luminescence, electroluminescence

(controlling exciton diffusion in multilayer white phosphorescent org. light emitting devices)

1T 4733-39-5, 2,9-Dimethyl-4,7-diphenyl-1,10-phenanthroline 50851-57-5, Poly(styrene sulfonic acid) 58328-31-7 123847-85-8 126213-51-2, Poly(3,4-ethylenedioxythiophene) 337526-88-2 343978-79-0 376367-93-0

(controlling exciton diffusion in multilayer white phosphorescent orq. light emitting devices)

IT 94928-86-6, Tris(2-phenylpyridine)iridium (controlling exciton diffusion in multilayer white phosphorescent org. light emitting devices)

L32 ANSWER 19 OF 29 HCA COPYRIGHT 2003 ACS on STN

136:86382 Electrochromic Properties of Laminate Devices
Fabricated from Polyaniline, Poly(ethylenedioxythiophene), and
Poly(N-methylpyrrole). Boehme, Jeffrey L.; Mudigonda, Dhurjati S.
K.; Ferraris, John P. (Department of Chemistry, The University of
Texas at Dallas, Richardson, TX, 75083-0688, USA). Chemistry of
Materials, 13(12), 4469-4472 (English) 2001. CODEN: CMATEX. ISSN:
0897-4756. Publisher: American Chemical Society.

The electrochromism of laminated conducting polymer layers AB was studied in a test assembly with vanadium pentoxide as counter-electrode an a polymer gel electrolyte. laminate systems were studied using combinations of poly(3,4-ethylenedioxythiophene), poly(N-methylpyrrole), and polyaniline as the active electrochromes. The conducting polymers were prepd. by electrochem. polymn. onto ITO substrates. electrolyte comprised poly(Me methacrylate), propylene carbonate, and ethylene carbonate in acetonitrile with lithium tetrafluoroborate and was spin-coated onto the V205 surface. The color change of the multilayers was measured using a spectrocolorimeter and analyzed using Commission Internationale de l'Eclairage 1931 (x, y)-chromaticity coordinates. The color of the conducting polymer layers in the fully oxidized and reduced states is linearly dependent on the color coordinates of the two individual polymers that comprise the laminate.

126213-51-2P, Poly(ethylenedioxythiophene)
(prepn. and tailored electrochromism of polyaniline and poly(ethylenedioxythiophene) and poly(N-methylpyrrole) conducting polymer multilayers)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CC 36-5 (Physical Properties of Synthetic High Polymers) Section cross-reference(s): 73, 76

ST polyaniline conducting polymer multilayer electrochromism chromaticity coordinate; ethylenedioxythiophene homopolymer laminate prepn electrochromism; methylpyrrole homopolymer multilayer electrochromism measurement; conducting polymer prepn electrochem polymn multilayer electrochromism

IT Polymerization

(electrochem.; prepn. and tailored electrochromism of polyaniline

```
and poly(ethylenedioxythiophene) and poly(N-methylpyrrole)
        conducting polymer multilayers)
IT
     Conducting polymers
        (polypyrroles; prepn. and tailored electrochromism of polyaniline
        and poly(ethylenedioxythiophene) and poly(N-methylpyrrole)
        conducting polymer multilayers)
TT
     Conducting polymers
        (polythiophenes; prepn. and tailored electrochromism of
        polyaniline and poly(ethylenedioxythiophene) and
        poly(N-methylpyrrole) conducting polymer multilayers)
IT
     Electrochromic materials
     Electrochromism
       Multilayers
        (prepn. and tailored electrochromism of polyaniline and
        poly(ethylenedioxythiophene) and poly(N-methylpyrrole) conducting
        polymer multilayers)
ΙT
     Polyanilines
        (prepn. and tailored electrochromism of polyaniline and
        poly(ethylenedioxythiophene) and poly(N-methylpyrrole) conducting
        polymer multilayers)
IT
     Coating process
        (spin; prepn. and tailored electrochromism of
        polyaniline and poly(ethylenedioxythiophene) and
        poly(N-methylpyrrole) conducting polymer multilayers)
IT
     1314-62-1, Vanadium oxide (V2O5), uses
        (counter-electrode; prepn. and tailored electrochromism of
        polyaniline and poly(ethylenedioxythiophene) and
        poly(N-methylpyrrole) conducting polymer multilayers)
     75-05-8, Acetonitrile, uses
                                    96-49-1, Ethylene carbonate
IT
     108-32-7, Propylene carbonate
                                      9011-14-7, Poly(methyl methacrylate)
     14283-07-9, Lithium tetrafluoroborate
        (polymer gel electrolyte; prepn. and tailored electrochromism of
        polyaniline and poly(ethylenedioxythiophene) and
        poly(N-methylpyrrole) conducting polymer multilayers)
     25233-30-1P, Polyaniline
                                 72945-66-5P, Poly(N-methylpyrrole)
IT
     126213-51-2P, Poly(ethylenedioxythiophene)
        (prepn. and tailored electrochromism of polyaniline and
        poly(ethylenedioxythiophene) and poly(N-methylpyrrole) conducting
        polymer multilayers)
     ANSWER 20 OF 29 HCA COPYRIGHT 2003 ACS on STN
L32
136:62438 Multilayer structures as stable hole-injecting
     electrodes for use in high efficiency organic electronic devices.
     Parker, Ian D.; Zhang, Chi (Uniax Corporation, USA). PCT Int. Appl.
     WO 2001099207 A2 20011227, 43 pp. DESIGNATED STATES: W: AE, AG,
     AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU,
     CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL,
     IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD,
     MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK,
     SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM; RW: AT, BE, BF, BJ, CF, CG, CH, CI, CM, CY,
     DE, DK, ES, FI, FR, GA, GB, GR, IE, IT, LU, MC, ML, MR, NE, NL, PT,
```

SE, SN, TD, TG, TR. (English). CODEN: PIXXD2. APPLICATION: WO 2001-US19482 20010618. PRIORITY: US 2000-PV212924 20000620.

AB Multilayer electrodes are described which comprise a first layer having a first layer cond., a second layer in contact with the first layer, the second layer comprising a conductive org. material having a second layer cond., and a third layer in contact with the second layer, the third layer comprising a conductive org. material having a third layer cond. greater than the second layer cond. and less than the first layer cond. Preferably, the second layer of the multilayer electrode comprises a blend of conjugated conductive org. polymer with nonconductive polymer. Pixellated displays and electronic devices, esp. devices including photoactive layers, employing the electrodes are also described.

IT 126213-51-2, Poly(ethylenedioxythiophene)

(multilayer structures as stable hole-injecting

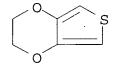
electrodes for use in high efficiency org. electronic devices)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



IC ICM H01L051-20

CC 76-2 (Electric Phenomena)

Section cross-reference(s): 38, 73, 74

ST multilayer hole injecting electrode org electronic device; display multilayer hole injecting electrode

IT Polyanilines

(doped; multilayer structures as stable hole-injecting electrodes for use in high efficiency org. electronic devices)

IT Electric contacts

Electrodes

(multilayer structures as stable hole-injecting

electrodes for use in high efficiency org. electronic devices)

IT Poly(arylenealkenylenes)

(multilayer structures as stable hole-injecting

electrodes for use in high efficiency org. electronic devices)

IT **Electroluminescent** devices

(org.; multilayer structures as stable hole-injecting

electrodes for use in high efficiency org. electronic devices)

IT 25233-30-1, Polyaniline

(doped; multilayer structures as stable hole-injecting

electrodes for use in high efficiency org. electronic devices)

26009-24-5, Poly(1,4-phenylenevinylene)

(multilayer structures as stable hole-injecting
electrodes for use in high efficiency org. electronic devices)

27119-07-9, Poly(2-acrylamido-2-methyl-1-propanesulfonic acid
(multilayer structures as stable hole-injecting
electrodes for use in high efficiency org. electronic devices)

9003-05-8, Poly(acrylamide) 126213-51-2,
Poly(ethylenedioxythiophene)
(multilayer structures as stable hole-injecting
electrodes for use in high efficiency org. electronic devices)

L32 ANSWER 21 OF 29 HCA COPYRIGHT 2003 ACS on STN

136:12471 Low-cost organic pulse sources for integrated optical modules.
Hiltunen, Jussi A.; Rantala, Juha T. (VTT Electronics, Oulu,
FIN-90570, Finland). Proceedings of SPIE-The International Society
for Optical Engineering, 4284 (Functional Integration of
Opto-Electro-Mechanical Devices and Systems), 108-114 (English)
2001. CODEN: PSISDG. ISSN: 0277-786X. Publisher: SPIE-The
International Society for Optical Engineering.

AB The transient and steady state performance of org. lightemitting devices (OLEDs) was studied with a view towards
suitability for pulse sources. The rise and fall times of the
electroluminescence of the different structures and

materials were afforded special attention.

cover single and multi-layer structures with different layer thicknesses. Both mol. and polymeric- based devices were tested. Mol. materials used in the OLEDs were N, N'-bis(3-methylphenyl)-N,N'-diphenylbenzidine (TPD) as a hole transporter, tris-(8-hydroxyquinolate) Al (Alq3) as an electron transporter/emitter and 4,7-diphenyl-1,10-phenanthroline (BCP) as a hole blocking material. Poly(2-methoxy, 5-(2'-ethyl-hexoxy)-1,4-phenylene-vinylene) (MEH-PPV) and poly(3,4-ethylenedioxythiophene)/poly(styrene) (PEDOT/PSS) were the polymeric

The tested devices

materials used in the devices. The effect of the driving voltage on the response time and the c.d. in transients was under study. In addn., changes in the response time were studied, when the bias voltage was applied.

IT 375846-91-6

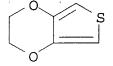
(low-cost org. pulse sources for integrated optical modules contg.)

RN 375846-91-6 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, polymer with ethenylbenzene (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



CM 2

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 38, 76

ST org light emitting device pulse source integrated optical module; delay time charge mobility bias voltage OLED

IT Luminescence, electroluminescence

(rise and fall times of **electroluminescence** of different structures and materials)

IT Multilayers

(tested devices cover single and multi-

layer structures with different layer thicknesses)

IT **Electroluminescent** devices

(thin-film; low-cost org. pulse sources for integrated optical modules in)

IT 375846-91-6

(low-cost org. pulse sources for integrated optical modules contq.)

L32 ANSWER 22 OF 29 HCA COPYRIGHT 2003 ACS on STN

135:324553 Effective design of blue organic electroluminescent devices by introducing functional monomeric layers. Choi, J.-H.; Jung, S.-H.; Kwon, S.-K.; Cho, W.-J.; Ha, C.-S. (Department of Polymer Science and Engineering, Pusan National University, Pusan, 609-735, S. Korea). Materials Science & Engineering, B: Solid-State Materials for Advanced Technology, B85(2-3), 96-99 (English) 2001. CODEN: MSBTEK. ISSN: 0921-5107. Publisher: Elsevier Science S.A..

AB Blue org. electroluminescent devices (OELDs), having a multi
-layered structure, were fabricated and their performance
was studied. A distyryl biphenyl arylene deriv. was synthesized as
a blue emitting material. To improve thermal stability of the
monomeric hole-transporting emissive material, poly(bisphenol
A-co-4-nitro phthalic anhydride-co-1,3-phenylene diamine) was used
as a matrix. For more effective design of the devices, poly(styrene
sulfonate) doped poly(3,4-ethylenedioxythiophene), and
2,9-dimethyl-4,7-diphenyl-1,10-phenanthroline (bathocuproine) and

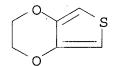
tris(8-quinolinolato)aluminum (Alq3) were introduced as a buffer layer, a hole-blocking layer, and an electron-injection layer, resp. The OELDs showed bright green color when the Bathocuproine layer was not applied.

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 22, 36, 76

IT Films

(multilayer; effective design of blue org. electroluminescent devices by introducing functional monomeric layers)

IT Coating process

(spin; effective design of blue org. electroluminescent devices by introducing functional monomeric layers)

IT 126213-51-2, Poly(3,4-ethylenedioxythiophene)
 (buffer layer, doped with poly(styrene sulfonate); effective
 design of blue org. electroluminescent devices by introducing
 functional monomeric layers)

L32 ANSWER 23 OF 29 HCA COPYRIGHT 2003 ACS on STN

135:280269 Electroluminescent devices employing organic luminescent material/clay nanocomposites. Park, O-Ok; Lee, Tae-Woo (Korea Advanced Institute of Science and Technology, S. Korea). PCT Int. Appl. WO 2001072925 Al 20011004, 20 pp. DESIGNATED STATES: W: DE, JP, KR, US. (English). CODEN: PIXXD2. APPLICATION: WO 2001-KR534 20010330. PRIORITY: KR 2000-16466 20000330.

AB Org. luminescent material/clay nanocomposites,
prepd. in a form of quantum well by blending an org.
luminescent material and a nanoclay, are described and
electroluminescent devices employing the nanocomposites as
active layers are discussed. The electroluminescent(
EL) devices comprise a transparent substrate; a
semitransparent electrode deposited on the transparent substrate; a

clay nanocomposite emissive layer spin-coated with the org. EL material/clay nanocomposite, positioned on the semitransparent electrode; and, a metal electrode deposited on the clay nanocomposite emissive layer. The EL devices can also contain a hole transporting layer positioned on the semitransparent electrode and/or an electron transporting layer positioned on the clay nanocomposite emissive layer. IT 126213-51-2, Polyethylene dioxythiophene (semitransparent electrode; electroluminescent devices employing org. luminescent material/clay nanocomposites contq.) 126213-51-2 HCA RN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX CN NAME) CM 1 126213-50-1 CRN CMF C6 H6 O2 S IC ICM C09K011-00 C09K011-06; H05B033-14 ICS 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties) Section cross-reference(s): 38, 76 electroluminescent device ora STluminescent clay nanocomposite; OLED polymer clay nanocomposite; luminescent material org polymer clay nanocomposite Amines, uses IT (aryl, tertiary, hole-transporting layer; electroluminescent devices employing org. luminescent material/clay nanocomposites contg.) ITLaminated materials (clay; electroluminescent devices employing org . luminescent material/clay nanocomposites contg.) Amines, uses IT (diamines, arom.; electroluminescent devices employing org. luminescent material/clay nanocomposites contq.) ITAlloys, uses (electrode; electroluminescent devices employing org. luminescent material/clay nanocomposites contq.) IT Electroluminescent devices

Luminescent substances

```
Nanocomposites
     Quantum well devices
        (electroluminescent devices employing org.
        luminescent material/clay nanocomposites)
IT
     Clays, uses
        (electroluminescent devices employing org.
        luminescent material/clay nanocomposites)
IT
     Glass substrates
        (electroluminescent devices employing org.
        luminescent material/clay nanocomposites contg.)
IT
     Coordination compounds
     Polyacetylenes, uses
     Polyanilines
     Polymers, uses
     Polyguinolines
        (electroluminescent devices employing org.
        luminescent material/clay nanocomposites contg.)
IT
     Poly(arylenealkenylenes)
        (poly(arylene vinylene); electroluminescent devices
        employing org. luminescent material/clay
        nanocomposites contq.)
     Polyquinoxalines
IT
        (polyphenylquinoxalines, poly(phenylquinoxaline);
        electroluminescent devices employing org.
        luminescent material/clay nanocomposites contg.)
IT
     Polyesters, uses
        (substrate; electroluminescent devices employing
        org. luminescent material/clay nanocomposites
     7439-89-6, Iron, uses 7439-92-1, Lead, uses 7439-93-2, Lithium,
IT
            7440-05-3, Palladium, uses 7440-06-4, Platinum, uses -4, Silver, uses 7440-33-7, Tungsten, uses 7440-50
     7440-22-4, Silver, uses
                                                             7440-50-8,
                    7440-57-5, Gold, uses 7440-66-6, Zinc, uses
     Copper, uses
     7440-74-6, Indium, uses
        (electrode; electroluminescent devices employing
        org. luminescent material/clay nanocomposites
        conta.)
                                  7440-70-2, Calcium, uses
IT
     7439-95-4, Magnesium, uses
        (electrode; electroluminescent devices employing
        org. luminescent material/clay nanocomposites
        contq.)
IT
     7429-90-5, Aluminum, properties
        (electrode; electroluminescent devices employing
        org. luminescent material/clay nanocomposites
        contq.)
IT
     9003-53-6, Polystyrene
        (electroluminescent devices employing org.
        luminescent material/clay nanocomposites contg.)
     25067-59-8, Poly(N-vinylcarbazole)
                                          115708-89-9
IT
        (electroluminescent devices employing org.
        luminescent material/clay nanocomposites contg.)
IT
     138184-36-8
```

```
(electroluminescent devices employing org.
        luminescent material/clay nanocomposites contq.)
                                                        517-51-1, Rubrene
IT
     120-12-7, Anthracene, uses
                                 198-55-0, Perylene
     7385-67-3, Nile red 25067-58-7, Polyacetylene
                                                        25087-26-7
                                     25233-34-5, Polythiophene
     25190-62-9, Poly(p-phenylene)
     30604-81-0, Polypyrrole 38215-36-0, coumarin 6
     4-(Dicyanomethylene)-2-methyl-6-(p-dimethylaminostyryl)-4H-pyran
     65181-78-4, (N,N'-Diphenyl-N,N'-bis(3-methylphenyl)-1,1'-biphenyl-
     4,4'-diamine)
                     95270-88-5, Polyfluorene
                                               150405-69-9
        (electroluminescent devices employing org.
        luminescent material/clay nanocomposites contg.)
IT
                   203915-07-5
                                 302921-88-6
     192198-85-9
        (electron-transporting layer; electroluminescent
        devices employing org. luminescent
        material/clay nanocomposites contg.)
IT
     2085-33-8, Alq3
        (electron-transporting layer; electroluminescent
        devices employing org. luminescent
        material/clay nanocomposites contg.)
IT
     288-13-1, Pyrazole
                          58328-31-7
                                       123847-85-8, 4,4'-Bis[N-(-naphthyl-
     1-)-N-phenylamino]biphenyl
        (hole-transporting layer; electroluminescent devices
        employing org. luminescent material/clay
        nanocomposites contq.)
     1318-93-0, Montmorillonite, properties
IT
        (nanoclay; electroluminescent devices employing
        org. luminescent material/clay nanocomposites
        contg.)
     1318-74-7, Kaolinite, uses
                                53320-86-8, Laponite
ΙT
        (nanoclay; electroluminescent devices employing
        org. luminescent material/clay nanocomposites
        contq.)
IT
     1335-25-7, Lead oxide 126213-51-2, Polyethylene
    dioxythiophene
        (semitransparent electrode; electroluminescent devices
        employing org. luminescent material/clay
        nanocomposites contg.)
IT
     50926-11-9, Indium tin oxide
        (semitransparent electrode; electroluminescent devices
        employing org. luminescent material/clay
        nanocomposites contg.)
                                25038-59-9, Polyethylene terephthalate,
ΙT
     14808-60-7, Quartz, uses
        (substrate; electroluminescent devices employing
        org. luminescent material/clay nanocomposites
        contg.)
     ANSWER 24 OF 29 HCA COPYRIGHT 2003 ACS on STN
L32
135:187082 White and blue temperature stable and efficient OLEDs using
     amorphous spiro transport and spiro emitting compounds. Spreitzer,
```

Hubert; Vestweber, Horst; Stoessel, Philipp; Becker, Heinrich

(Covion Organic Semiconductors GmbH, Frankfurt, D-65926, Germany).

Proceedings of SPIE-The International Society for Optical Engineering, 4105 (Organic Light-Emitting Materials and Devices IV), 125-133 (English) 2001. CODEN: PSISDG. ISSN: 0277-786X. Publisher: SPIE-The International Society for Optical Engineering. The temp. stability of white and blue OLEDs was studied by observing AΒ the I-V, EL-V and the spectral characteristics of various devices stored at elevated temp. (.ltoreg.130.degree.). Blue multilayer org. light emitting diodes (OLEDs) contg. PEDOT (polyethylenedioxythiophene) or PANI (polyaniline) derivs. as the hole injection and buffer layer, arom. diamines like Spiro-TAD (2,2',7,7'-tetrakis(diphenylamino)spiro-9,9'bifluorene) as a hole transport material (HTM), Spiro-DPVBi (2,2',7,7'-tetrakis(2,2-diphenylvinyl)spiro-9,9'-bifluorene) as an emitting material (EM) and of Alq3 (tris(8hydroxyquinolinato) aluminum) as the electron-injection and electron-transport layer (ETL) were fabricated. White OLEDs were prepd., contg. an addnl. DCM (dicyanmethylene-2-methyl-6-(pdimethylaminostyryl)-4H-pyran) doped Alq3 layer between the Spiro-DPVBi and Alq3 layer. Use of Spiro-TAD as a hole transport material (HTM) and of Spiro-DPVBi as an emitting material (EM) resulted in dramatically improved temp. stability: for the white and blue OLED no significant deterioration up to 130.degree. Devices consisting of non spiro components like NPB were found. and/or DPVBi already started to degrade at much lower temps. 126213-51-2, Polyethylenedioxythiophene

IT

(white and blue temp. stable and efficient LEDs using amorphous spiro transport and spiro emitting compds. and)

RN126213-51-2 HCA

Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX CN NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

73-5 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties)

Section cross-reference(s): 76

IT Electric current-potential relationship

Electric transport properties

Luminescence, electroluminescence

(of white and blue temp. stable and efficient LEDs using amorphous spiro transport and spiro emitting compds.)

IT Electroluminescent devices

(white and blue temp. stable and efficient LEDs using amorphous

spiro transport and spiro emitting compds.) IT

2085-33-8, Tris(8-hydroxyquinolinato)aluminum 25233-30-1,

Polyaniline 124729-98-2, MTDATA **126213-51-2**,

Polyethylenedioxythiophene

(white and blue temp. stable and efficient LEDs using amorphous spiro transport and spiro emitting compds. and)

ANSWER 25 OF 29 HCA COPYRIGHT 2003 ACS on STN L32

135:172719 Transient electroluminescence in multilayer organic light-emitting diodes: experiment and

theory. Book, K.; Nikitenko, V. R.; Bassler, H.; Elschner, A. (Institute of Physical, Macromolecular and Nuclear Chemistry,

Philipps University, Marburg, D-35032, Germany). Synthetic Metals, 122(1), 135-139 (English) 2001. CODEN: SYMEDZ. ISSN: 0379-6779.

Publisher: Elsevier Science S.A..

The authors have studied a multilayer org. light AΒ

-emitting diode (OLED) with 1,3,5-tris(N,N-bis-(4,5-methoxy-phenyl)-aminophenyl)-benzene (TAPB) acting as hole transporting layer (HTL) and tris(8-hydroxy-quinolinolato) Al (Alq3) as electron transporting layer (ETL). Pos. charge carriers in the HTL were detected optically as a function of the applied bias. Also, the authors studied the DC-characteristics of current and brightness as well as the onset behavior of the electroluminescence (EL) as a function of the applied bias. An anal. model is developed to describe the obsd.

carrier concns. as well as the current-voltage characteristics and the transient EL measurements quant.

IT 126213-51-2, PEDOT

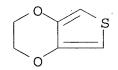
> (transient electroluminescence in multilayer org. light-emitting diodes)

RN 126213-51-2 HCA

Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX CNNAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



73-11 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties)

transient electroluminescence multilayer org ST

light emitting diode

IT Electric current-potential relationship

Electroluminescent devices

Luminescence, electroluminescence

(transient electroluminescence in multilayer

org. light-emitting diodes)

IT 2085-33-8, Aluminum Tris(8-hydroxyquinolinato) 50851-57-5 126213-51-2, PEDOT 142894-38-0

(transient electroluminescence in multilayer org. light-emitting diodes)

L32 ANSWER 26 OF 29 HCA COPYRIGHT 2003 ACS on STN

134:319491 Transparent electrically conductive film involving polythiophene layer. Tateno, Katsutaka; Yamada, Hiroyuki (Oji Paper Co., Ltd., Japan). Jpn. Kokai Tokkyo Koho JP 2001113635 A2 20010424, 5 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 1999-301020 19991022.

AB The film involves a 10-500-nm transparent elec. conductive film on .gtoreq.1 side of a transparent polymer film support and a polythiophene-type transparent elec. conductive polymer layer on the above conductive film. The film, showing good bending resistance and prevention of coloration, is suitable for a liq. crystal display device, etc.

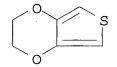
126213-51-2, 3,4-Polyethylenedioxythiophene
(polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



IC ICM B32B027-00

ICS B32B007-02; C23C014-06; C23C014-08; H01B005-14

CC 76-2 (Electric Phenomena)

Section cross-reference(s): 38, 75

IT Polyesters, uses

(arom.; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT Transparent films

(elec. conductive; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT Electric conductors

(films, transparent; polymer substrate-supported laminate

of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT Vapor deposition process

(ion plating; polymer substrate-supported laminate of ion-plated transparent elec. conductor film and polythiophene-type conductive film)

IT Polysulfones, uses

(polyether-, support; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT Polyethers, uses

(polysulfone-, support; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT Polymers, uses

(polythiophenes; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT Heat-resistant materials

Transparent films

(support; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT Polycarbonates, uses

Polyesters, uses

(support; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

IT 50926-11-9, ITO

(film; polymer substrate-supported **laminate** of ion-plated transparent elec. conductor film and polythiophene-type conductive film)

IT 126213-51-2, 3,4-Polyethylenedioxythiophene (polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)

- IT 9020-73-9, Poly(ethylene naphthalate) 24968-11-4, Poly(ethylene naphthalate) 25038-59-9, PET (polyester), uses (support; polymer substrate-supported laminate of transparent elec. conductor film and polythiophene-type conductive film with bending resistance)
- L32 ANSWER 27 OF 29 HCA COPYRIGHT 2003 ACS on STN
- 134:272953 Surface roughness effects and their influence on the degradation of organic light emitting devices. Jonda, Ch.; Mayer, A. B. R.; Stolz, U.; Elschner, A.; Karbach, A. (Corporate Research and Development, Robert Bosch GmbH, Gerlingen, D-70839, Germany). Journal of Materials Science, 35(22), 5645-5651 (English) 2000. CODEN: JMTSAS. ISSN: 0022-2461. Publisher: Kluwer Academic Publishers.
- AB Org. light emitting devices typically consist of one or several org. layers which are sandwiched between

two electrodes, one of which has to be transparent. In most cases In Sn oxide (ITO) is employed as the transparent, hole-injecting anode material. Usually, the functional org. layers possess a thickness of .apprx.100 nm. For such thin films the homogeneity and the surface roughness are esp. important factors for the device performance. Therefore, the surface roughness of all those layers which are the basis for subsequent deposition processes were systematically studied by at. force microscopy (AFM). For these studies both the ITO substrate and the layers consisting of different org. materials deposited onto the ITO substrate were analyzed. In addn., the two different basic deposition methods for the org. materials, the deposition from soln. by spin coating and the deposition by thermal evapn., were compared to one another with respect to their resulting surface roughness. The large surface roughness of the ITO substrate induces layer inhomogeneities, esp. for the vapor deposited org. layers. They can be reduced by the incorporation of a polymeric smoothing layer.

IT 126213-51-2, Poly(3,4-ethylenedioxythiophene)

(surface roughness effects and influence on degrdn. of org. light emitting devices)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 38

IT Coating process

(spin; surface roughness effects and influence on degrdn. of org. light emitting devices)

IT 2085-33-8, Aluminum tris(8-hydroxyquinolinato) 50851-57-5
126213-51-2, Poly(3,4-ethylenedioxythiophene) 128366-35-8
(surface roughness effects and influence on degrdn. of org. light emitting devices)

L32 ANSWER 28 OF 29 HCA COPYRIGHT 2003 ACS on STN

131:293118 Flexible substrates for organic device. Burroughes, Jeremy Henley; Devine, Peter (Cambridge Display Technology Limited, UK). Eur. Pat. Appl. EP 949850 A1 19991013, 10 pp. DESIGNATED STATES: R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO. (English). CODEN: EPXXDW. APPLICATION: EP

1999-302512 19990331. PRIORITY: GB 1998-7149 19980402.

(substantially) transparent formable and/or flexible component for use as an outer protective element in an electronic or optoelectronic device including .gtoreq.1 elec. active org. layer are described in which the component is a composite structure comprising a layer of glass .ltoreq.200 .mu.m thick and a layer of plastic. Org. devices, esp. org. light-emitting devices, are described which employ the protective composites. Methods for fabricating the devices are also described which entail forming the composite and laminating it to a device, forming the composite and forming the device on it, or forming the device structure on a plastic layer and then laminating the plastic to a glass layer.

IT 126213-51-2D, polystyrene sulfonic acid-doped

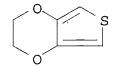
(glass-plastic composites in flexible substrates for org. devices and the devices and their fabrication)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S



IC ICM H05B033-02

ICS H05B033-04; H05B033-10; H01L051-20

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

ST protective composite flexible substrate org device; composite glass plastic substrate flexible org device; **OLED** composite glass plastic substrate; **electroluminescent** device composite glass plastic substrate

IT Electric apparatus

Electroluminescent devices

Semiconductor devices

(org.; glass-plastic composites in flexible substrates for org. devices and the devices and their fabrication)

IT 50926-11-9, Indium tin oxide 96638-49-2, Poly(phenylene vinylene) 126213-51-2D, polystyrene sulfonic acid-doped 138184-36-8, MEH-PPV

(glass-plastic composites in flexible substrates for org. devices and the devices and their fabrication)

L32 ANSWER 29 OF 29 HCA COPYRIGHT 2003 ACS on STN

131:177292 Advances in Ch-LCD devices using plastic substrates with conducting polymer. Fritz, William J.; Wonderly, H.; Smith, Steven W.; Kim, Y.; Chonko, J.; Doane, J. William; Shashidhar, Ranganathan; O'Ferrall, Catherine Elizabeth; Cuttino, David S. (Liquid Crystal Institute, Kent State Univ., Kent, OH, USA). Proceedings of SPIE-The International Society for Optical Engineering, 3635(Liquid Crystal Materials, Devices, and Applications VII), 114-119 (English) 1999. CODEN: PSISDG. ISSN: 0277-786X. Publisher: SPIE-The International Society for Optical Engineering.

Cholesteric liq. crystal display (Ch-LCD) are lightwt., low power, AB sunlight readable displays. In addn., they can serve a dual function as pen-input device switch no addnl. hardware. Because of the unique properties of this technol., Ch-LCDs can be made with plastic substrates thus making the displayed extremely lightwt., compact and unbreakable. We discuss in this paper recent advances in merging Ch-LCD technol. with conducting polymer electrodes. Conducting polymer provides potential benefits over the use of the std. display electrode materials, indium tin oxide, by improving the reliability of the display. Furthermore, the potential to print the conducting polymer electrodes could significantly increase manufg. vol. and decrease display cost. We report on scaling display size and resoln. by demonstrating a 1/8 VGA Ch-LCD using polypyrrole as the conducting polymer. We fabricated these displays using either a vacuum fill or polymer wall/lamination approach and we discus subsequent failure anal. to det. the cause for the line-outs obsd. on these displays. We present initial results in detg. the suitability for using Ch-LCD technol. as a pen-input device. Finally, we discuss initial work towards printing the conducting polymer electrodes to det. the feasibility of printing electrodes on plastic substrates in a roll-to-roll, high vol., low cost process. IT

(conducting polymer; advances in cholesteric liq. crystal display devices using plastic substrates with conducting polymer)

RN 126213-51-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 38

IT Conducting polymers
Liquid crystal displays
(advances in choleste

(advances in cholesteric liq. crystal display

devices using plastic substrates with conducting polymer)

IT Liquid crystals

(cholesteric; advances in cholesteric liq. crystal display devices using plastic substrates with conducting polymer)

IT Polyesters, uses

(substrate; advances in cholesteric liq. crystal **display devices** using plastic substrates with conducting polymer)

IT 30604-81-0, Polypyrrole **126213-51-2**

(conducting polymer; advances in cholesteric liq. crystal display devices using plastic substrates with conducting polymer)

IT 25038-59-9, uses

(substrate; advances in cholesteric liq. crystal **display devices** using plastic substrates with conducting polymer)

=> d his 136-

FILE 'HCA' ENTERED AT 12:38:44 ON 30 JUL 2003 E OPTICAL IMAGING DEVICES/CV

L36 29420 S E3

L37 4 S L36 AND L10

L38 3 S L37 NOT (L24 OR L32)

=> d l38 1-3 cbib abs hitstr hitind

L38 ANSWER 1 OF 3 HCA COPYRIGHT 2003 ACS on STN

138:63944 Electrochromic organic polymer synthesis and devices utilizing electrochromic organic polymers. Xu, Chunye; Taya, Minoru (University of Washington, USA). U.S. Pat. Appl. Publ. US

2002196518 A1 20021226, 29 pp., which which (English). CODEN:

USXXCO. APPLICATION: US 2002-180222 20020625. PRIORITY: US

2001-PV300675 20010625; US 2001-PV324205 20010921; US 2002-PV364418

Laminated electrochromic devices are described which comprise a transparent electrode layer; a cathodic polymer (e.g., poly[3,3-dimethyl-3,4-dihydro-2H-thieno[3,4-b][1,4]dioxepine]) layer; an electrolyte layer comprising a solid electrolyte; and a counter electrode layer. An anodic polymer layer (e.g., poly(3,6-bis(2-(3,4-ethylenedioxythiophene))-N-methylcarbazole)) may be formed on the electrolyte layer under the counter electrode. Surface plasmon resonance imaging systems, electrochromic windows, and electrochromic diplays employing the devices are also described. Methods for prepg. the cathodic polymer are described which entail reflux of a toluene soln. of 3,4-dimethoxythiophene and 2,2-dimethyl-1,3-propanediol to produce a monomer precursor. Methods for producing the anodic polymer are also described which

entail first producing intermediates and then reacting the intermediates.

IT 177038-66-3P 255901-53-2P

(org. polymer electrochromic devices and electrochromic polymer synthesis)

RN 177038-66-3 HCA

CN Poly[(9-methyl-9H-carbazole-3,6-diyl)(2,2',3,3'-tetrahydro[5,5'-bithieno[3,4-b]-1,4-dioxin]-7,7'-diyl)] (9CI) (CA INDEX NAME)

RN 255901-53-2 HCA

CN 2H-Thieno[3,4-b][1,4]dioxepin, 3,4-dihydro-3,3-dimethyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 255901-50-9 CMF C9 H12 O2 S

IC ICM G02F001-03

ICS G02F001-07

NCL 359245000

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 38, 72

IT Optical imaging devices

(surface plasmon resonance; using org. polymer electrochromic devices)

- IT 177038-66-3P 255901-53-2P
 - (org. polymer electrochromic devices and electrochromic polymer synthesis)
- L38 ANSWER 2 OF 3 HCA COPYRIGHT 2003 ACS on STN
- 133:59821 Electroconductive glass laminate. Cloots, Tom;
 Louwet, Frank; Andriessen, Hieronymus; Verlinden, Bart; Tahon,
 Jean-Pierre; Vermeulen, Leo; Leenders, Luc; Goedeweeck, Rudi
 (Agfa-Gevaert N.V., Belg.). Eur. Pat. Appl. EP 1013413 A1 20000628,
 11 pp. DESIGNATED STATES: R: AT, BE, CH, DE, DK, ES, FR, GB, GR,
 IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO. (English).
 CODEN: EPXXDW. APPLICATION: EP 1998-204382 19981221.
- AB A material is disclosed which comprises a substrate and an org. electroconductive layer provided on said substrate, characterized in that the substrate is a laminate comprising a glass layer and a support. The glass layer is preferably a flexible glass layer having a thickness from 10 to 500 .mu.m. The material can be used as an electrode in elec. or semiconductor devices thereby providing an improved lifetime, e.g. Displays, photovoltaic cells or light-emitting diodes.
- IC ICM B32B017-10 ICS C03C027-12
- CC 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 76
- ST electroconductive glass laminate polythiophene
- IT Electric conductors

Electroluminescent devices

Optical imaging devices

Photoelectric devices

(electroconductive glass laminate)

- IT Borosilicate glasses
 - Glass, uses

(electroconductive glass laminate)

- IT 25233-34-5P, Poly-thiophene **126213-51-2P**,
 - 3,4-Ethylenedioxy-thiophene homopolymer (electroconductive glass laminate)
- L38 ANSWER 3 OF 3 HCA COPYRIGHT 2003 ACS on STN
- 130:58888 Conductive layer system and use thereof in electroluminescent systems. Huppauff, Martin; Sybrichs, Ralf; Gehrig, Andreas (Robert Bosch Gmbh, Germany). PCT Int. Appl. WO 9854767 A1 19981203, 21 pp. DESIGNATED STATES: W: JP, US; RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LÜ, MC, NL, PT, SE. (German). CODEN: PIXXD2. APPLICATION: WO 1998-DE1467 19980529. PRIORITY: DE 1997-19722946 19970531; DE 1997-19757874 19971224.
- AB Transparent or semitransparent conductive layer systems consisting of org. and inorg. elec. conductive materials are described which comprise .gtoreq.2 layers, the first layer contg. an org. or organometallic elec. conductive polymer which is transparent or semitransparent in the visible range of the electromagnetic spectrum while the second contains at least one elec. conductive inorg. compd. or a metal or a metalloid doped accordingly. The layer

systems forms a multilayer hybrid electrode for use as a cathode in electroluminescent systems. Use in displays is indicated. 126213-51-2, 3,4-Polyethylenedioxythiophene IT (elec. conductive multilayered systems including org. and inorq. layers and their use in electroluminescent systems) RN 126213-51-2 HCA Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, homopolymer (9CI) (CA INDEX CN NAME) CM 1 CRN 126213-50-1 CMF C6 H6 O2 S IC ICM H01L051-20 ICS H05B033-28 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties) Section cross-reference(s): 76 Cathodes IT Electric contacts Electroluminescent devices Optical imaging devices (elec. conductive multilayered systems including org. and inorg. layers and their use in electroluminescent systems) ITPolyacetylenes, uses Polyanilines (elec. conductive multilayered systems including org. and inorg. layers and their use in electroluminescent systems) IT Polymers, uses (polythiophenes; elec. conductive multilayered systems including org. and inorg. layers and their use in electroluminescent systems) Aluminum alloy IT Chromium alloy Copper alloy Gold alloy Iron alloy Palladium alloy Platinum alloy Silver alloy Tin alloy (elec. conductive multilayered systems including org. and inorg. layers and their use in electroluminescent systems) 7429-90-5, Aluminum, uses 7439-89-6, Iron, uses 7440-05-3, IT

7440-06-4, Platinum, uses

Palladium, uses

7440-22-4, Silver,

7440-31-5, Tin, uses 7440-44-0, Carbon, uses 7440-47-3, 7440-50-8, Copper, uses 7440-57-5, Gold, uses Chromium, uses 25233-30-1, Polyaniline 25233-34-5, Polythiophene 11099-20-0 30604-81-0, Polypyrrole 30604-81-0D, Polypyrrole, derivs. 50926-11-9, Indium tin oxide **126213-51-2**, 3,4-Polyethylenedioxythiophene (elec. conductive multilayered systems including org. and inorq. layers and their use in electroluminescent systems) 7429-90-5, Aluminum, uses 7439-89-6, Iron, uses IT 7440-05-3, 7440-06-4, Platinum, uses 7440-22-4, Silver, Palladium, uses 7440-31-5, Tin, uses 7440-44-0, Carbon, uses 7440-47-3, 7440-50-8, Copper, uses 7440-57-5, Gold, uses Chromium, uses 25233-30-1, Polyaniline 25233-34-5, Polythiophene 11099-20-0 30604-81-0D, Polypyrrole, derivs. 30604-81-0, Polypyrrole 50926-11-9, Indium tin oxide 126213-51-2, 3,4-Polyethylenedioxythiophene (elec. conductive multilayered systems including org. and inorg. layers and their use in electroluminescent systems) => d his 139-FILE 'REGISTRY' ENTERED AT 12:51:18 ON 30 JUL 2003 1 S 126213-51-2 L39 FILE 'HCA' ENTERED AT 12:51:47 ON 30 JUL 2003 1148 S L39 L4010 S L33 NOT L40 L41 => d 141 1-10 cbib abs hitstr hitind ANSWER 1 OF 10 HCA COPYRIGHT 2003 ACS on STN 139:77169 Design of a screen printable electrode for an organic light-emitting device. Carter, Sue A.; Victor, John (Add-Vision, Inc., USA). PCT Int. Appl. WO 2003054981 A1 20030703, 27 pp. DESIGNATED STATES: W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM; RW: AT, BE, BF, BJ, CF, CG, CH, CI, CM, CY, DE, DK, ES, FI, FR, GA, GB, GR, IE, IT, LU, MC, ML, MR, NE, NL, PT, SE, SN, TD, TG, TR. (English). CODEN: PIXXD2. APPLICATION: WO 2002-US41353 20021220. PRIORITY: US 2001-PV342579 20011220. AΒ The invention relates to the design of a screen printable electrode for an org. light emitting device. An electroluminescent device consists of a plurality of layers, where the plurality of layers

includes (i) a bottom electrode layer; (ii) a light-

emitting material layer, such that the lightemitting material layer is created over the bottom electrode
layer; and (iii) a top electrode layer, such that the top electrode
layer is printed under atm. conditions over the lightemitting material layer.

IT 332951-15-2, 3,4-Ethylenedioxythiophene-styrenesulfonic acid copolymer .

(conductive paste contg.; design of a screen printable electrode for an org. light-emitting device)

RN 332951-15-2 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-, polymer with ethenylbenzene monosulfo deriv. (9CI) (CA INDEX NAME)

CM 1

CRN 126213-50-1 CMF C6 H6 O2 S

CM 2

CRN 30105-09-0 CMF C8 H8 O3 S CCI IDS

 $H_2C = CH - Ph$

D1-SO3H

IC ICM H01L051-20 ICS H01L051-40

CC 76-2 (Electric Phenomena)
Section cross-reference(s): 38, 66, 73, 74

ST screen printable electrode org light emitting device

IT Metals, uses

(composite, conductive paste contg.; design of a screen printable electrode for an org. **light-emitting** device)

IT Conducting polymers

(conductive paste contg.; design of a screen printable electrode for an org. light-emitting device)

IT Bromides, uses Chlorides, uses ΙT

IT

IT

IT

IT

IT

IT

IT

IT

Fluorides, uses Halides Iodides, uses Oxides (inorganic), uses Polyanilines Polymers, uses Salts, uses Sulfates, uses (conductive paste contg.; design of a screen printable electrode for an org. light-emitting device) Polymers, uses (conjugated, electroluminescent material; design of a screen printable electrode for an org. lightemitting device) Electric contacts Electrically conductive pastes Electroluminescent devices Ink-jet printing Printing (impact) Printing (nonimpact) Screen printing (design of a screen printable electrode for an org. light -emitting device) (elec. conductive; design of a screen printable electrode for an org. light-emitting device) Sol-gel processing (electrode layer contq.; design of a screen printable electrode for an org. light-emitting device) Luminescent substances (electroluminescent, films; design of a screen printable electrode for an org. light-emitting device) Electric conductors (films; design of a screen printable electrode for an org. light-emitting device) Surfactants (ionic, electrode layer contg.; design of a screen printable electrode for an org. light-emitting device) Esters, uses (solvent; design of a screen printable electrode for an org. light-emitting device) 51-92-3D, Tetramethylammonium, salts 62-53-3D, Phenylamine, salts 66-40-0D, Tetraethylammonium, salts 76-05-1D, Trifluoroacetic 104-15-4D, Toluenesulfonic acid, salts 603-34-9, acid, salts 1493-13-6D, Trifluoromethylsulfonic acid, salts Triphenylamine 7429-90-5D, Aluminum, salts 7439-93-2D, Lithium, salts 7440-02-0, Nickel, uses 7440-09-7D, Potassium, salts 7440-22-4, Silver, uses 7440-23-5D, Sodium, salts 7440-39-3D, Barium, salts 7440-44-0, Carbon, uses 7440-46-2D, Cesium, salts 7440-70-2D, 10549-76-5D, Tetrabutylammonium, salts Calcium, salts 13010-31-6D, Tetrapropylammonium, salts 15477-33-5, Aluminum

chlorate 16872-11-0D, Tetrafluoroboric acid, salts 16940-81-1D, Hexafluorophosphoric acid, salts 25233-30-1, Polyaniline 33906-65-9D, Borate(1-), tetraphenyl-, hydrogen, salts 332951-15-2, 3,4-Ethylenedioxythiophene-styrenesulfonic acid copolymer

(conductive paste contg.; design of a screen printable electrode for an org. light-emitting device)

51-92-3D, Tetramethylammonium, salts 62-53-3D, Phenylamine, salts IT 66-40-0D, Tetraethylammonium, salts 76-05-1D, Trifluoroacetic 104-15-4D, Toluenesulfonic acid, salts acid, salts 603-34-9, Triphenylamine 1493-13-6D, Trifluoromethylsulfonic acid, salts 7429-90-5D, Aluminum, salts 7439-93-2D, Lithium, salts 7440-02-0, Nickel, uses 7440-09-7D, Potassium, salts 7440-22-4, 7440-39-3D, Barium, salts 7440-23-5D, Sodium, salts Silver, uses 7440-46-2D, Cesium, salts 7440-70-2D, 7440-44-0, Carbon, uses 10549-76-5D, Tetrabutylammonium, salts Calcium, salts 13010-31-6D, Tetrapropylammonium, salts 15477-33-5, Aluminum 16872-11-0D, Tetrafluoroboric acid, salts 16940-81-1D, chlorate Hexafluorophosphoric acid, salts 25233-30-1, Polyaniline 33906-65-9D, Borate(1-), tetraphenyl-, hydrogen, salts 332951-15-2, 3,4-Ethylenedioxythiophene-styrenesulfonic acid copolymer

(conductive paste contg.; design of a screen printable electrode for an org. light-emitting device)

- L41 ANSWER 2 OF 10 HCA COPYRIGHT 2003 ACS on STN
- 139:60536 Transfer material of organic thin-film device and manufacture of organic thin-film device by using the same. Tateishi, Tomomi (Fuji Photo Film Co., Ltd., Japan). Jpn. Kokai Tokkyo Koho JP 2003178868 A2 20030627, 12 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 2001-334858 20011031. PRIORITY: JP 2001-305429 20011001.
- AB The transfer material consists of a temporary support having thereon .gtoreq.1 org. thin-film layers which are to be transfered to a substrate by laminating and heating and/or pressurizing, the surface roughness of the temporal support has the max. height Rmax (JIS B 0601-1982) .ltoreq.50 per 100 of the thickness of the org. thin-film layer, thereby offering good interfacial adhesion between the transfered org. thin-film layer and a device substrate. Preferably, the org. thin-film layer contains at least a luminescent compd. and./or a carrier-transporting compd. In the manuf. of the org. thin-film device, and from the substrate side, a hole-transporting org. thin-film layer, a luminescent org. thin-film layer, and an electron-transporting org. thin-film layer are transfered in this order. The substrate may consist of a substrate support having thereon a transparent elec. conductive film.
- IT **155090-83-8**, Baytron P

(hole-transporting layer; manuf. of org. **El** device by using transfer material composed of org. thin-film device supported on temporal support)

RN 155090-83-8 HCA

CN Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-51-2 CMF (C6 H6 O2 S)x CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S

CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



 $D1-CH=CH_2$

D1-SO3H

IC ICM H05B033-10 ICS H05B033-14

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

- ST org thin film device manuf transfer material; transfer film org electroluminescent device
- IT **Electroluminescent** devices

(manuf. of org. El device by using transfer material composed of org. thin-film device supported on temporal support)

- IT Polycarbonates, uses Polyesters, uses

(temporal support; manuf. of org. **El** device by using transfer material composed of org. thin-film device supported on temporal support)

- 1T 94928-86-6, Tris(2-phenylpyridine)iridium (luminescent layer; manuf. of org. El device by using transfer material composed of org. thin-film device supported on temporal support)

- L41 ANSWER 3 OF 10 HCA COPYRIGHT 2003 ACS on STN
- 137:330855 Production methods of organic **electroluminescent** devices. Kobe, Emiko (Tdk Corporation, Japan). Jpn. Kokai Tokkyo Koho JP 2002324670 A2 20021108, 7 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 2001-126368 20010424.
- The devices comprise: a lower laminate contg. a polymer electroluminescent phosphor layer; an upper laminate coated using an org. solvent selected from the following: (1) a mol. contg. a C3-6 chain compd. having a C2-3 alkoxy (and carbonyl) group and a C2-3 ester compd., both of which contain OH at an .alpha. and/or a .beta. position; (2) a mol. having a C3-6 chain compd. contg. a C2-4 dialkyl amide group; (3) a C5-8 chain compd. ester; and (4) a C4-7 chain compd. carbonate.

```
IT
     155090-83-8, Baytron P
        (prodn. methods of org. electroluminescent devices)
RN
     155090-83-8 HCA
     Benzenesulfonic acid, ethenyl-, homopolymer, compd. with
CN
     2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX
     NAME)
     CM
          1 .
     CRN
          126213-51-2
     CMF
          (C6 H6 O2 S)x
     CCI
          PMS
          CM
```

CRN

CMF

126213-50-1

C6 H6 O2 S

CM3 CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI **PMS** CM4 CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



 $D1-CH=CH_2$

 $D1-SO_3H$

IC ICM H05B033-10

ICS C09K011-06; H05B033-14; H05B033-22 CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties) polymer org electroluminescent device STIT Solvents (org.; prodn. methods of org. electroluminescent devices) IT Amide group Carbonyl group Chemical chains Electroluminescent devices Glass substrates Hydroxyl group Laminated materials Phosphors (prodn. methods of org. electroluminescent devices) IT Polymers, uses (prodn. methods of orq. electroluminescent devices) IT Carbonates, reactions (prodn. methods of org. electroluminescent devices) IT Esters, reactions (prodn. methods of org. electroluminescent devices) IT 123-86-4, n-Butyl acetate 2085-33-8, Tris(8-quinolinolato)aluminum 50926-11-9, ITO **155090-83-8**, Baytron P 37271-44-6 188201-14-1 444716-92-1 473799-92-7 (prodn. methods of orq. electroluminescent devices) L41ANSWER 4 OF 10 HCA COPYRIGHT 2003 ACS on STN 137:192553 Organic electroluminescent devices using thermoplastic substrates and their manufacture. Mishima, Masayuki (Fuji Photo Film Co., Ltd., Japan). Jpn. Kokai Tokkyo Koho JP 2002246172 A2 20020830, 9 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 2001-37501 20010214. AΒ The org. EL device has a thermoplastic substrate having thereon transparent electrodes, .gtoreq.1 org. compd. layers involving luminescent layers, back electrodes, and a thermoplastic sealing which seals the org. compd. layer(s) and shields outside airs and is fused with the thermoplastic substrate around the periphery of the luminescent laminate to offer excellent brightness, luminescent efficiency and durability. The device is useful for full color displays, back lights, surface light sources, light source arrays for printers, etc. 155090-83-8, Baytron P ΙT (hole injection layer; manuf. of org. EL devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

Benzenesulfonic acid, ethenyl-, homopolymer, compd. with

2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX

CM 1

NAME)

RN

CN

155090-83-8

HCA

```
CRN 126213-51-2

CMF (C6 H6 O2 S) x

CCI PMS

CM 2

CRN 126213-50-1

CMF C6 H6 O2 S
```

CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



 $D1-CH=CH_2$

 $D1-SO_3H$

IC ICM H05B033-04

ICS H05B033-02; H05B033-10; H05B033-14

- CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
- org electroluminescent device thermoplastic substrate durability; sealing thermoplastic substrate org electroluminescent device
- IT Fluoropolymers, uses (Nitoflon, substrate; manuf. of org. EL devices using

thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT Polycarbonates, uses

(Panlite, substrate; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT Polyesters, uses

(Tetoron Film, substrate; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT Sealing

(manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT **Electroluminescent** devices

(org.; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT Plastics, uses

(thermoplastics, substrates; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT 117944-65-7, Indium zinc oxide

(IZO, transparent electrode; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT 7440-22-4, Silver, uses 12614-86-7

(Mg-Ag/Ag laminate back electrode; manuf. of org. EL devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT 1312-43-2, Indium oxide (In2O3)

(Zn-doped In2O3 transparent electrode; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT 7440-66-6, Zinc, uses

(dopant, Zn-doped In2O3 transparent electrode; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT 358974-66-0, 2,2',2''-(1,3,5-Benzenetriyl)tris[3-(2-methylphenyl)-3H-imidazo[4,5-b]pyridine]

(electron transporting layer; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT 15082-28-7, 2-(4-Biphenylyl)-5-(4-tert-butylphenyl)-1,3,4-oxadiazole 123847-85-8

(electron-transporting material; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

IT 155090-83-8, Baytron P

(hole injection layer; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)

- IT 58328-31-7, 4,4'-N,N'-Dicarbazolylbiphenyl (host material; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)
- IT 58328-31-7, 4,4'-N,N'-Dicarbazolylbiphenyl (host material; manuf. of org. **EL** devices using thermoplastic substrates sealed with thermoplastic sealings for enhanced durability)
- L41 ANSWER 5 OF 10 HCA COPYRIGHT 2003 ACS on STN
- 137:176899 Macromolecular **electroluminescent** element and its production method. Shimizu, Takao; Iguchi, Mayumi; Sekine, Tokumasa; Minato, Takao (Toppan Printing Co., Ltd., Japan). Jpn Kokai Tokkyo Koho JP 2002231444 A2 20020816, 6 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 2001-21186 20010130.
- AB The invention refers to a prodn. method of a macromol.

 electroluminescent element, wherein some polymer layers are
 laminated onto an anode substrate comprising a metallic
 layer to form the first bonding surface, and the remaining polymer
 layers are formed on a transparent cathode substrate to form the 2nd
 bonding surface, and two surfaces are bonded together, wherein the
 1st bonding surface is treated to have av. surface roughness of 0.05
 10 .mu.m, in order to prevent outside light reflections and to
 form an electroluminescent element with good bonding.
- IC ICM H05B033-10 ICS H05B033-14
- CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
- ST electroluminescent device bonding surface roughness
- IT Adhesive bonding

Electroluminescent devices

Surface roughness

(macromol. electroluminescent element and prodn.
method)

- IT 138184-36-8, MEH-PPV 155090-83-8 (macromol. electroluminescent element and prodn. method)
- L41 ANSWER 6 OF 10 HCA COPYRIGHT 2003 ACS on STN 137:39135 Method for producing electroluminescent element.

Kashiwabara, Mitsuhiro (Japan). U.S. Pat. Appl. Publ. US 20020072139 A1 20020613, 24 pp. (English). CODEN: USXXCO. APPLICATION: US 2001-960088 20010921. PRIORITY: JP 2000-289946 20000925; JP 2001-156787 20010525; JP 2001-164212 20010531; JP 2001-169143 20010605.

AB Methods for producing electroluminescent elements are described in which .gtoreq.1 org. electroluminescent layer constituting the electroluminescent element is patterned

by the use of a photolithog. method. Electroluminescent elements are also described which comprise .gtoreq.1 patterned org. electroluminescent layer, wherein the electroluminescent element does not have any one of a partition, a structure aiding patterning, and surface treatment aiding patterning. Preferably, the width of an area with uneven film thickness that is formed at the end of the patterned luminous layer is .ltoreq.15 .mu.m. The devices may comprise plural patterned luminous layers that can emit light with different colors, and the distance between the adjacent luminous layers emitting different colors is .ltoreq.30 .mu.m.

IT **155090-83-8**, Baytron P

(org. electroluminescent device fabrication using photolithog. patterning of the org. layers and the devices)

RN 155090-83-8 HCA

Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

CN

CRN 126213-51-2 CMF (C6 H6 O2 S)x CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S

CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



 $D1-CH=CH_2$

 $D1-SO_3H$

IC ICM H01L021-00 438029000 NCL 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties) Section cross-reference(s): 74, 76 electroluminescent device fabrication photolithog ST patterning Semiconductor device fabrication IT (electroluminescent devices; org. electroluminescent device fabrication using photolithog. patterning of the org. layers and the devices) IT Photolithography (in org. electroluminescent device fabrication) Electroluminescent devices IT (orq. electroluminescent device fabrication using photolithog. patterning of the org. layers and the devices)

(org. electroluminescent device fabrication using photolithog. patterning of the org. layers and the devices)

L41 ANSWER 7 OF 10 HCA COPYRIGHT 2003 ACS on STN
136:361648 Polymer electroluminescent (EL) elements
with improved mechanical strength. Sekine, Tokumasa; Kai, Teruhiko;
Shimizu, Takao (Toppan Printing Co., Ltd., Japan). Jpn. Kokai
Tokkyo Koho JP 2002141174 A2 20020517, 5 pp. (Japanese). CODEN:
JKXXAF. APPLICATION: JP 2000-334255 20001101.

AB The EL element has a polymer lightemitting layer between 2 electrodes (at least one of which is transparent or translucent), wherein the electrode(s) is formed on a thermoplastic resin adhesive layer.

IT 155090-83-8, Baytron P (hole-transporting layer; polymer EL elements having

electrodes **laminated** via thermoplastic resin adhesives for improving mech. strength)

RN 155090-83-8 HCA

Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

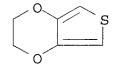
CN

CRN 126213-51-2 CMF (C6 H6 O2 S)x

CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S



CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



D1-CH=CH2

D1-SO3H

IC ICM H05B033-26

ICS H05B033-10; H05B033-14

- CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
 - Section cross-reference(s): 38
- ST polymer thin film electroluminescent element strength; polypropylene adhesive lamination aluminum EL element; polyphenylenevinylene light emitting layer EL device
- IT Poly(arylenealkenylenes)

(phosphor layer; polymer **EL** elements having electrodes **laminated** via thermoplastic resin adhesives for improving mech. strength)

IT Adhesives

(polymer EL elements having electrodes laminated via thermoplastic resin adhesives for improving mech. strength)

IT Plastics, uses

(thermoplastics, adhesive layer; polymer **EL** elements having electrodes **laminated** via thermoplastic resin adhesives for improving mech. strength)

IT **Electroluminescent** devices

(thin-film; polymer **EL** elements having electrodes **laminated** via thermoplastic resin adhesives for improving mech. strength)

IT 9002-88-4D, Polyethylene, acid-modified 9003-07-0D, Polypropylene, acid-modified 24937-78-8, Ethylene-vinyl acetate copolymer 129408-58-8D, Adtex ER 523L, acid-modified 263015-28-7, Admer QE 060

(adhesive layer; polymer **EL** elements having electrodes **laminated** via thermoplastic resin adhesives for improving mech. strength)

IT **155090-83-8**, Baytron P

(hole-transporting layer; polymer **EL** elements having electrodes **laminated** via thermoplastic resin adhesives for improving mech. strength)

IT 155090-83-8, Baytron P

(hole-transporting layer; polymer **EL** elements having electrodes **laminated** via thermoplastic resin adhesives for improving mech. strength)

- L41 ANSWER 8 OF 10 HCA COPYRIGHT 2003 ACS on STN
- 136:361630 Electroluminescent (EL) elements sealed with gas- and moisture-barrier laminates with improved durability. Arai, Koji; Tsuzuki, Atsuo (Dainippon Printing Co., Ltd., Japan). Jpn. Kokai Tokkyo Koho JP 2002134271 A2 20020510, 9 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 2000-330031 20001030.
- The element contg. a 1st electrode, an **EL** layer, and a 2nd electrode, useful for a display, is sealed with a **laminate** of .gtoreq.2 barrier substrates having a base layer and a coating layer, which is preferably formed by PVD or CVD of an inorg. oxide. The **laminate** may have a water absorbent-contg. adhesive

layer between the substrate.

155090-83-8, Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (Baytron PTP AI 4083, electrode; durable EL elements sealed with gas- and moisture-barrier laminates)

RN 155090-83-8 HCA

CN Benzenesulfonic acid, ethenyl-, homopolymer, compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 126213-51-2 CMF (C6 H6 O2 S)x CCI PMS

CM 2

CRN 126213-50-1 CMF C6 H6 O2 S

CM 3

CRN 50851-57-5 CMF (C8 H8 O3 S)x CCI PMS

CM 4

CRN 26914-43-2 CMF C8 H8 O3 S CCI IDS



```
D1-CH=CH_2
```

D1-SO3H

```
IC
     ICM H05B033-04
         H05B033-10; H05B033-14
CC
     73-11 (Optical, Electron, and Mass Spectroscopy and Other Related
     Properties)
    Section cross-reference(s): 38
     electroluminescent element durability gas barrier
ST
     laminate; silica deposition PET laminate sealing
    EL; org EL display barrier laminate
     sealant
IT
     Polyesters, uses
        (arom., barrier laminate; durable EL elements
        sealed with gas- and moisture-barrier laminates)
    Oxides (inorganic), uses
IT
        (barrier coat layer; durable EL elements sealed with
        qas- and moisture-barrier laminates)
    Fluoropolymers, uses
IT
    Polyamides, uses
    Polycarbonates, uses
    Polyesters, uses
    Polyimides, uses
     Polyoxymethylenes, uses
    Polysiloxanes, uses
    Polysulfones, uses
    Polythiophenylenes
    Polyurethanes, uses
        (barrier laminate; durable EL elements sealed
        with gas- and moisture-barrier laminates)
TT
     Polyolefins
        (cyclic, barrier laminate; durable EL
        elements sealed with gas- and moisture-barrier laminates
     Electroluminescent devices
IT
        (durable EL elements sealed with gas- and
        moisture-barrier laminates)
IT
     Acetals
        (polyacetals, nonpolymeric, barrier laminate; durable
        EL elements sealed with gas- and moisture-barrier
```

laminates) IT Polyimides, uses (polyamide-, barrier laminate; durable EL elements sealed with gas- and moisture-barrier laminates IT Polysulfones, uses (polyether-, barrier laminate; durable EL elements sealed with gas- and moisture-barrier laminates IT Polyamides, uses (polyimide-, barrier laminate; durable EL elements sealed with gas- and moisture-barrier laminates IT Polyethers, uses (polysulfone-, barrier laminate; durable EL elements sealed with gas- and moisture-barrier laminates 155090-83-8, Benzenesulfonic acid, ethenyl-, homopolymer, IT compd. with 2,3-dihydrothieno[3,4-b]-1,4-dioxin homopolymer (Baytron PTP AI 4083, electrode; durable EL elements sealed with gas- and moisture-barrier laminates) IT418755-56-3P, 2,7-Bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-9,9-dihexylfluorene-2,7-dibromo-9,9-dioctylfluorene copolymer (EL layer; durable EL elements sealed with gas- and moisture-barrier laminates) 75-01-4D, Vinyl chloride, polymers 79-10-7D, Acrylic acid, IT 79-41-4D, Methacrylic acid, polymers 100-42-5D, 9003-54-7, Acrylonitrile-styrene copolymer Styrene, polymers 9003-56-9, ABS resin 9004-34-6D, Cellulose, derivs. Poly(ethylene terephthalate), uses (barrier laminate; durable EL elements sealed with gas- and moisture-barrier laminates) ΙT 7429-90-5, Aluminum, uses 7789-24-4, Lithium fluoride, uses (electrode; durable EL elements sealed with gas- and moisture-barrier laminates) 123863-97-8P, 9,9-Dihexylfluorene IT 189367-54-2P, 2,7-Dibromo-9,9-dihexylfluorene 254755-24-3P, 2,7-Bis(4,4,5,5tetramethyl-1,3,2-dioxaborolan-2-yl)-9,9-dihexylfluorene (for polyfluorene deriv. prepn.; durable EL elements sealed with gas- and moisture-barrier laminates) 86-73-7, Fluorene 111-25-1, n-Hexyl bromide 61676-62-8, IT 2-Isopropoxy-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (for polyfluorene deriv. prepn.; durable EL elements

ANSWER 9 OF 10 HCA COPYRIGHT 2003 ACS on STN 135:264430 Chemically amplified soft lithography of a low band gap polymer. Yu, Jianfei; Holdcroft, Steven (Department of Chemistry, Simon Fraser University, Burnaby, BC, V5A 1S6, Can.). Chemical Communications (Cambridge, United Kingdom) (14), 1274-1275 (English)

2001. CODEN: CHCOFS. ISSN: 1359-7345. Publisher: Royal

sealed with gas- and moisture-barrier laminates)

Society of Chemistry.

AB A solid state, acid-catalyzed reaction leading to chem. amplified soft lithog. is demonstrated with a low band gap conjugated polymer; poly({3-[11-(tetrahydropyran-2-yloxy)undecyl]-2,5-thiophenediyl}-3,4-ethylenedioxy-2,5-thiophenediyl). Chem. amplified soft lithog. is a non-photolithog. method that circumvents photochem. damage. Films are formed prior to patterning which may allow for further control of film thickness, morphol. and adhesion over other deposition methods. Since the patterned polymer is rendered insol., it is possible to deposit multiple layers of similar or dissimilar conjugated polymers. Evaluation of patterned films in field effect transistors, light-emitting and electrochromic devices is in progress.

IT 361432-87-3

(acid-catalyzed elimination of dihydropyran from low band-gap conjugated thiophene polymer)

RN 361432-87-3 HCA

CN 3-Thiopheneundecanol, 2-(2,3-dihydrothieno[3,4-b]-1,4-dioxin-5-yl)-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 361432-86-2 CMF C21 H30 O3 S2

IT 361432-85-1P

(chem. amplified soft lithog. of low band-gap thiophene polymer)

RN 361432-85-1 HCA

CN Thieno[3,4-b]-1,4-dioxin, 2,3-dihydro-5-[3-[11-[(tetrahydro-2H-pyran-2-yl)oxy]undecyl]-2-thienyl]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 361432-84-0 CMF C26 H38 O4 S2

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 76

IT 110-87-2 **361432-87-3**

(acid-catalyzed elimination of dihydropyran from low band-gap conjugated thiophene polymer)

IT 361432-85-1P

(chem. amplified soft lithog. of low band-gap thiophene polymer)

L41 ANSWER 10 OF 10 HCA COPYRIGHT 2003 ACS on STN

135:242592 Optical and electrochemical properties of soluble
N-hexylcarbazole-co-3,4-ethylenedioxythiophene copolymers. Beouch,
L.; Tran Van, F.; Stephan, O.; Vial, J. C.; Chevrot, C. (Equipe
Reactivite aux Interfaces (EA 2528), Laboratoire sur les Polymeres
et les Materiaux Electroactifs, Universite de Cergy Pontoise, Cergy
Pontoise, 95013, Fr.). Synthetic Metals, 122(2), 351-358 (English)
2001. CODEN: SYMEDZ. ISSN: 0379-6779. Publisher: Elsevier
Science S.A..

Sol. N-hexylcarbazole-co-3,4-ethylenedioxythiophene (HCz-co-EDOT) AB copolymers from mixts. in various ratio of the two corresponding dihalogenated monomers were synthesized. The random copolymers were obtained from EDOT/HCz starting molar ratio: 0.25, 1, 4, named, resp., C1, C2 and C3, and compared their properties to the two homopolymers synthesized in the same way. The IR studies clearly indicated, that an increase in the amt. of ethylenedioxythiophene in the feed compn. leads to an increase of the proportion of the corresponding comonomer in the final materials. Elemental anal. point out that the reactivity of dibrominated EDOT seems slightly lower than that of dibrominated HCz. Thin films of copolymer have been prepd. and their electrochem. response have been investigated. Absorption and luminescence of these materials have been also studied in CHCl3. Copolymers mainly composed of one monomer (C1 and C3) behaves like the corresponding homopolymers. On an other hand, copolymer (C2) obtained from an equimolar amt. of each monomer in the feed compn. clearly exhibits distinct signals in optical spectra

and in electrochem. behavior, probably due to the presence of each monomer unit short segments. The use of C2 has been explored for possible application in **light emitting** devices indicating that the p-doping of the material would be facilitated leading to an improved hole injecting when compared to carbazole homopolymer. It could be particularly interesting as a hole transporting layer in **multilayer** org. **light emitting** devices.

IT 163359-60-2P 350037-71-7P 359829-16-6P

(optical and electrochem. properties of sol. N-hexylcarbazole-co-3,4-ethylenedioxythiophene copolymers)

RN 163359-60-2 HCA

CN Poly(2,3-dihydrothieno[3,4-b]-1,4-dioxin-5,7-diyl) (9CI) (CA INDEX NAME)

RN 350037-71-7 HCA

CN Thieno[3,4-b]-1,4-dioxin, 5,7-dibromo-2,3-dihydro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 174508-31-7 CMF C6 H4 Br2 O2 S

RN 359829-16-6 HCA

CN 9H-Carbazole, 3,6-dibromo-9-hexyl-, polymer with 5,7-dibromo-2,3-dihydrothieno[3,4-b]-1,4-dioxin (9CI) (CA INDEX NAME)

CM 1



CRN 174508-31-7 CMF C6 H4 Br2 O2 S

CM 2

CRN 150623-72-6 CMF C18 H19 Br2 N

CC 35-5 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 73, 76

IT Absorption spectra Cyclic voltammetry Fluorescence

Hole transport

Luminescence

Luminescence, electroluminescence

(optical and electrochem. properties of sol. N-hexylcarbazole-co-3,4-ethylenedioxythiophene copolymers) \cdot

IT 163359-60-2P 350037-71-7P 359829-15-5P

359829-16-6P 359829-17-7P, Poly(9-hexyl-9H-carbazole-3,6-diyl)

(optical and electrochem. properties of sol. N-hexylcarbazole-co-3,4-ethylenedioxythiophene copolymers)